



Tailor-made Carbon Molecular Sieve Membranes (CMSMs) for gas separation in membrane reactors

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Inorganic Membranes
& Membrane Reactors

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Outlook

- Introduction
- CMSMs
- Applications of CMSMs
- Q&A

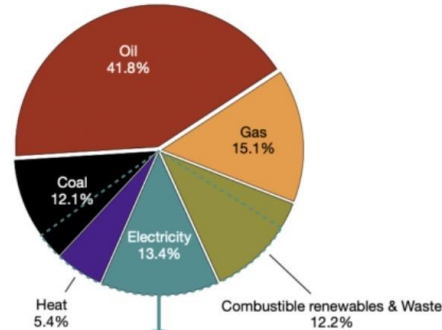


Introduction

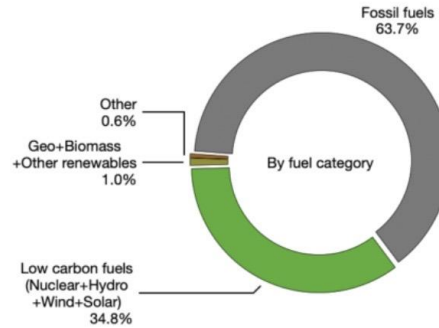
Energy resources

- The change in 30 years
- Dominated by fossil fuels (61%)
- Increased electricity production

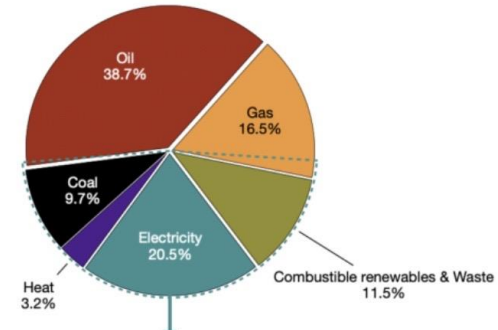
World Final Energy by share of fuel in 1990



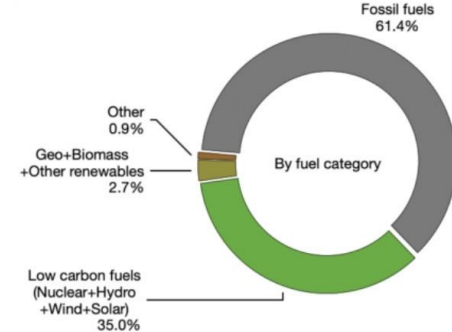
World Electricity Generation in 1990



World Final Energy by share of fuel in 2020
(Most recent year of data at time of publishing)



World Electricity Generation in 2021 (most recent)



<https://www.iea.org/data-and-statistics/data-product/world-energy-balances-highlights>

Introduction

SEPARATION OF CHEMICALS

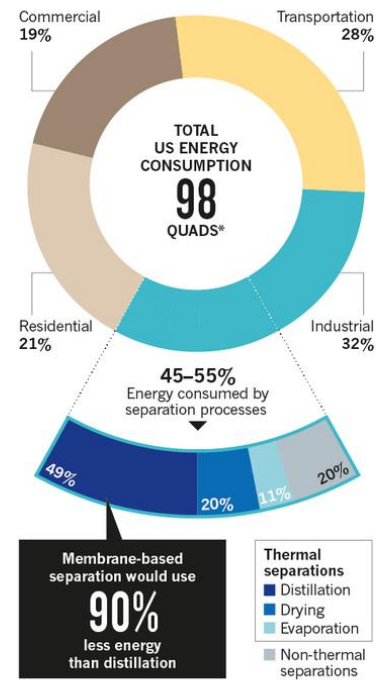
- 50% of US industrial energy use
- 10-15% of total energy consumption
- New technologies are required



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CUTTING COSTS

Chemical separations account for about half of US industrial energy use and 10-15% of the nation's total energy consumption. Developing alternatives that don't use heat could make 80% of these separations 10 times more energy efficient.



*A quad is a unit of energy equal to 10¹⁵ British Thermal Units (1 BTU is about 0.0003 kilowatt-hours).

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Introduction

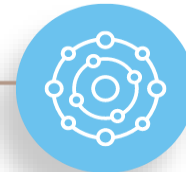
MEMBRANE SEPARATION TECHNOLOGY

- Carbon Molecular Sieve Membranes (CMSMs)
- Separation based on molecular **size** and/or **affinity**
- Sub-nanometer range

~100.000 times
smaller than human
hair thickness

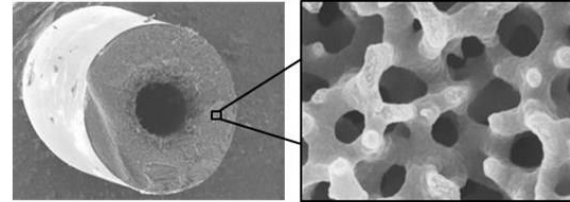
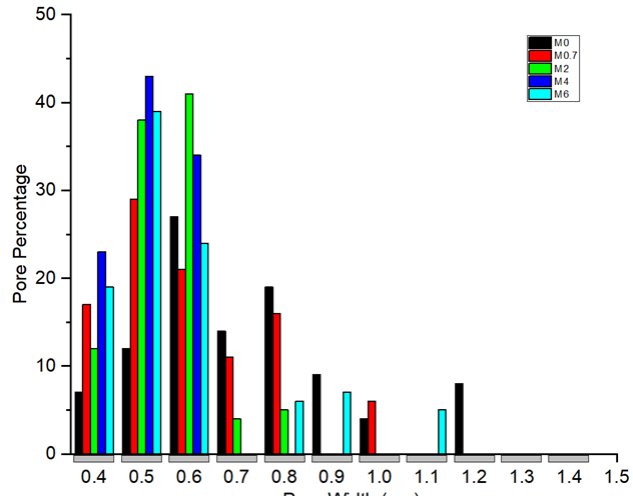


REQUIREMENTS



CMSMs

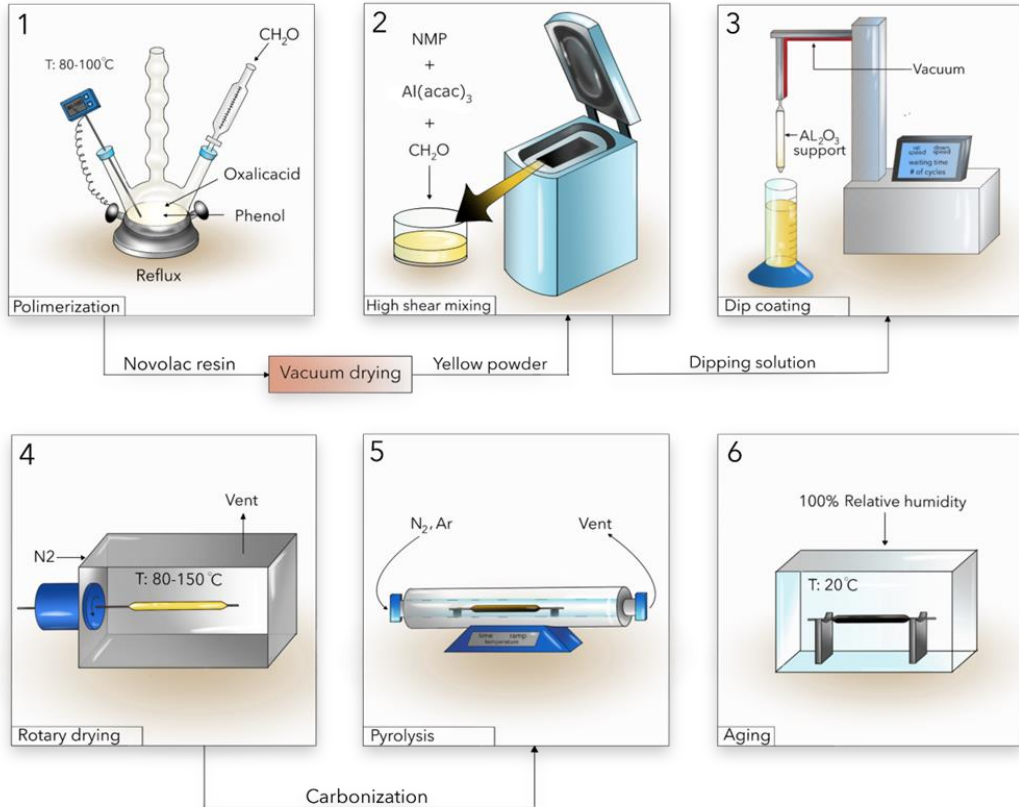
- Made via pyrolysis of a thermosetting polymer
 - Resorcinol- formaldehyde resin
 - Novolac
 - Polyimide based ...
- Porous structure
- Modification of pore size and pore size distribution



CMSMs

How is it made?

1. Precursor synthesis
2. Dipping solution preparation
3. Coating
4. Polymerization
5. Carbonization
6. Post treatment



Surface diffusion and molecular sieving are the common transport mechanisms

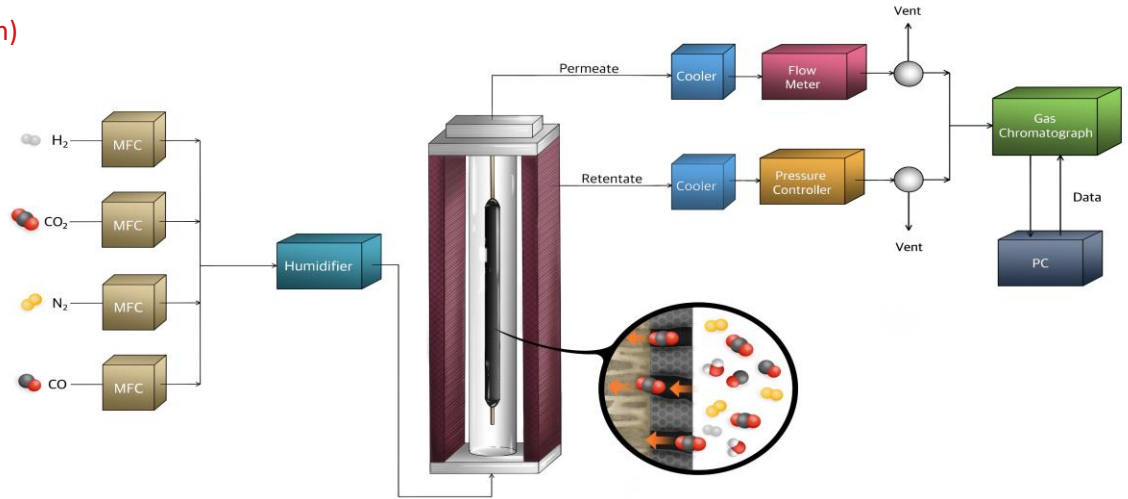
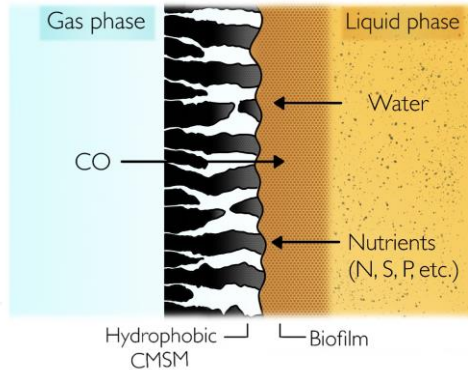
CMSMs

What could be separated?

Any desired molecule with **different kinetic diameter** compared to other molecules and/or **difference in polarity** (hydrophilicity/hydrophobicity)

Processes which have:

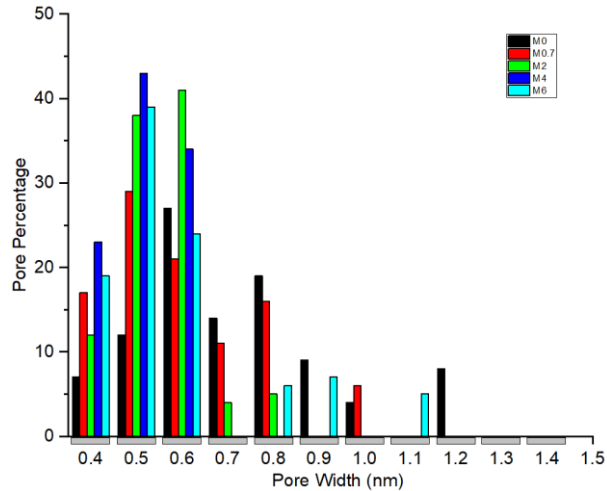
- Operational temperatures up to 700 °C (no oxygen)
- Operational pressures up to 200 Bar
- Configurations of
 - Gas separation
 - Pervaporation (liquid/vapor)
 - Vapor permeation
 - Selective gas distribution



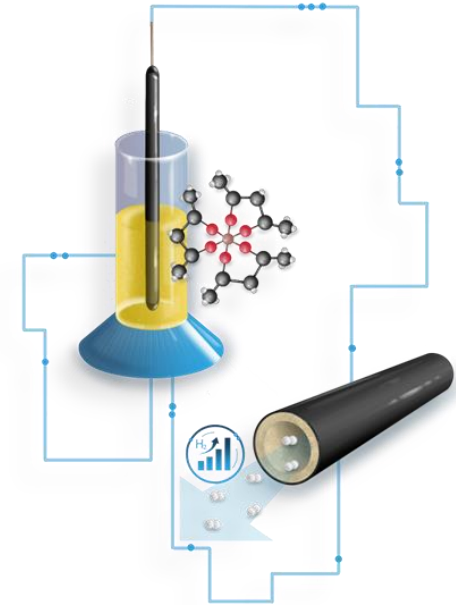
Applications CMSMs

H₂ separation

Al(acac)₃ effect on pore size distribution (PSD)



Narrow pore size distribution (PSD)



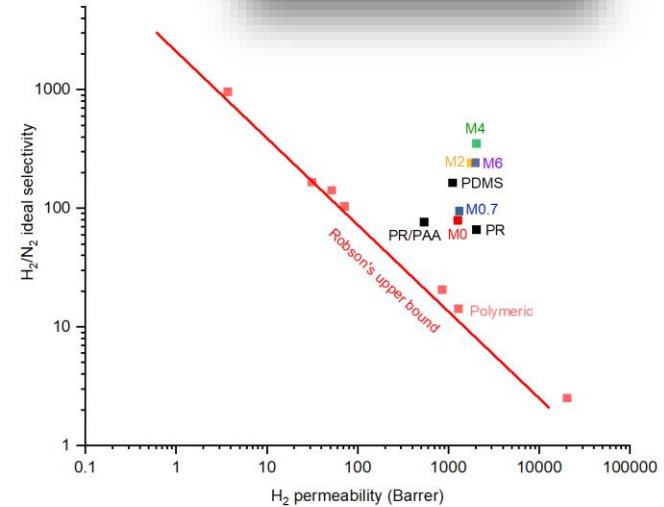
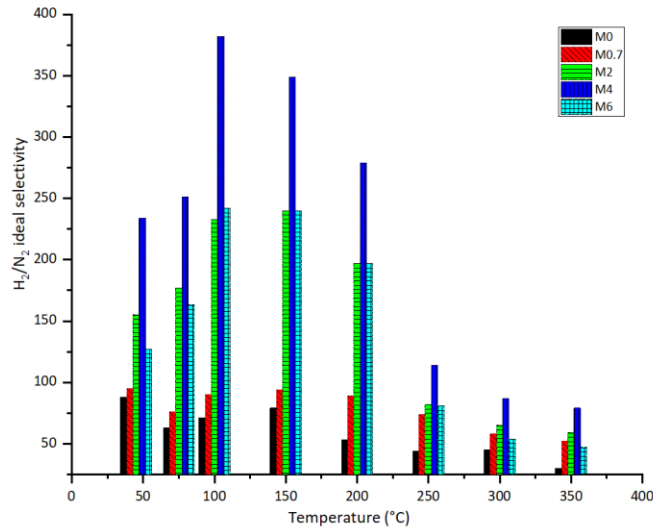
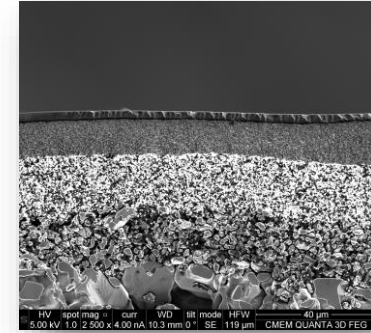
<https://doi.org/10.1016/j.ijhydene.2022.02.198>

Applications CMSMs

H₂ separation

Al(acac)₃ effect on pore size distribution (PSD)

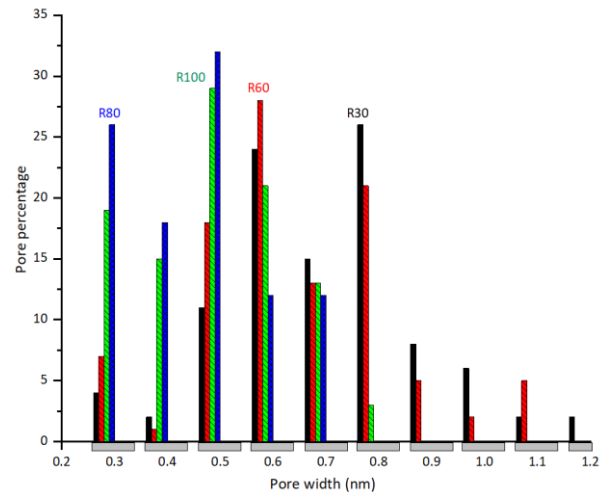
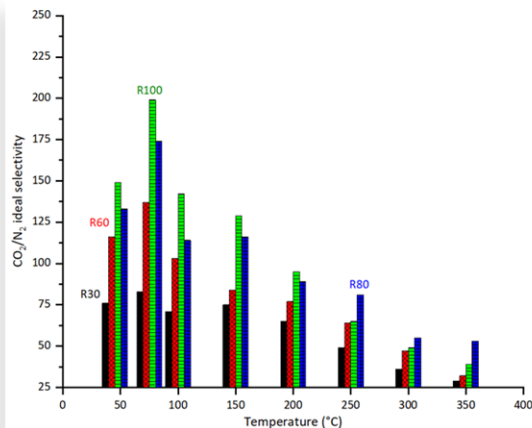
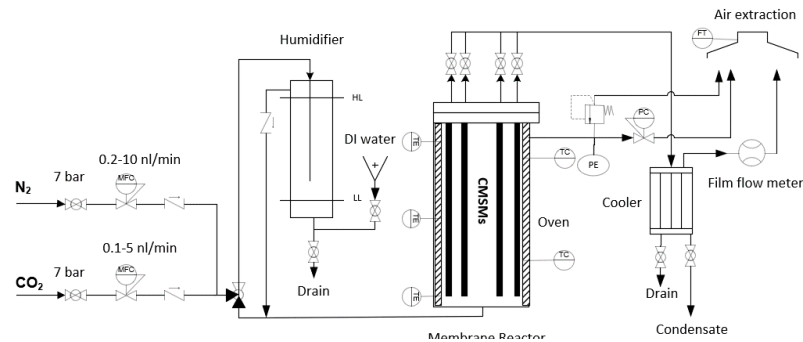
- 382 H₂/N₂ ideal selectivity
- Stable performance up to 350 °C



Applications CMSMs

CO₂ separation CO₂/N₂

Effect of degree of polymerization on PSD
Resorcinol-formaldehyde resin

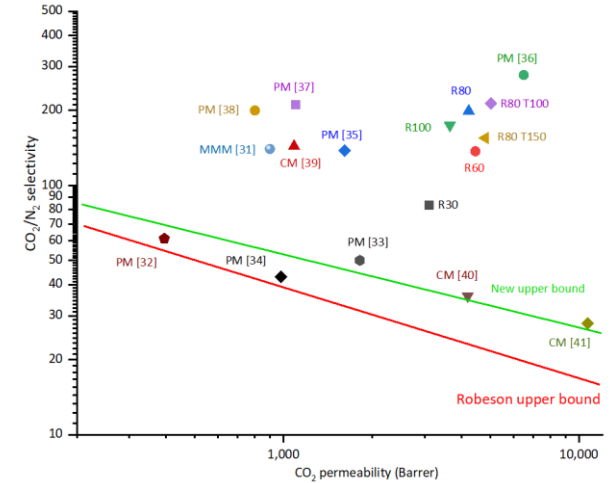
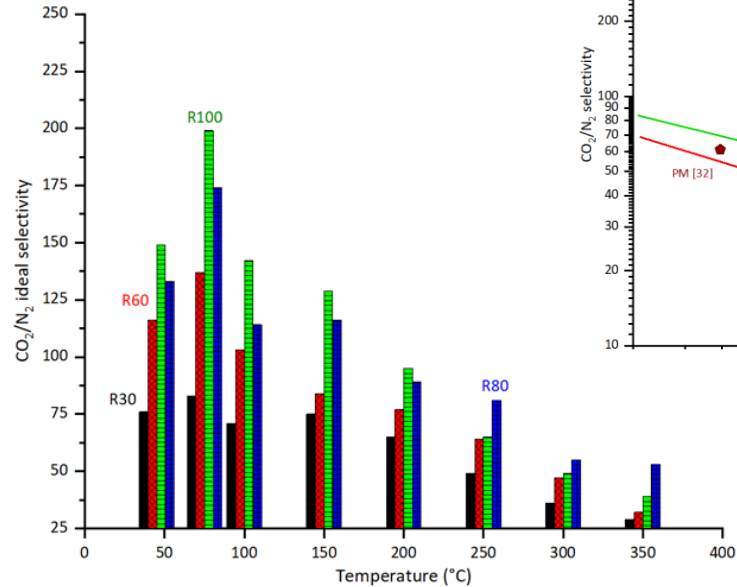
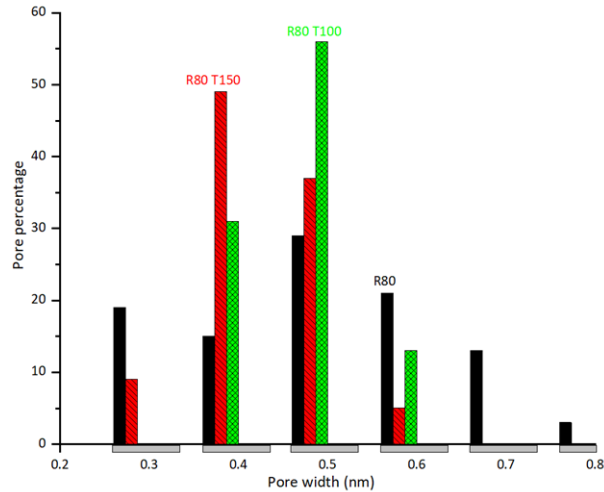


<https://doi.org/10.3390/membranes12090847>

Applications CMSMs

CO₂ separation CO₂/N₂

Post treatment with oxygen effect on PSD



Fine tuning the PSD with degree of polymerization and oxygen post treatment

<https://doi.org/10.3390/membranes12090847>



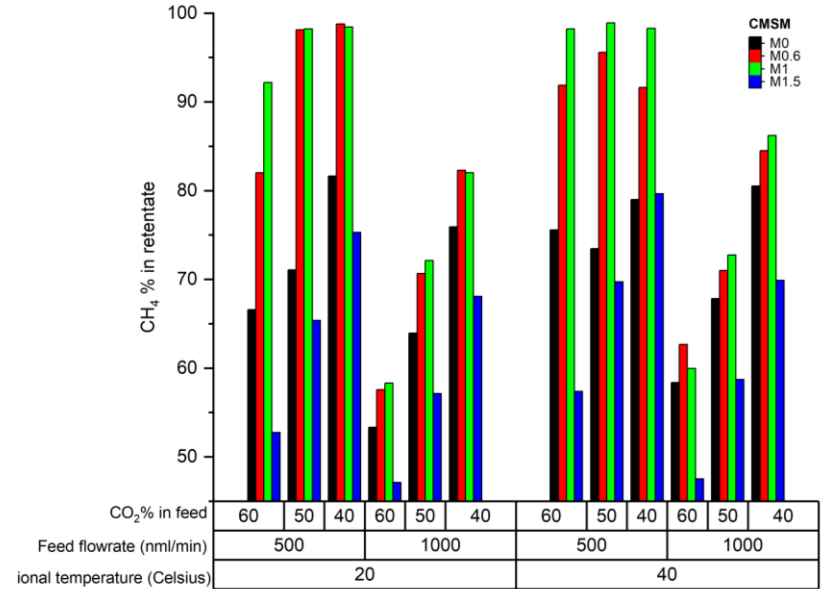
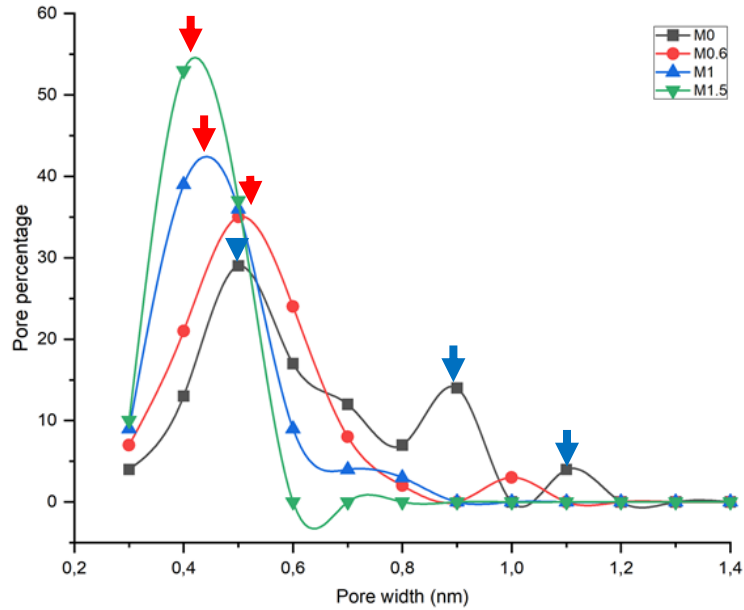
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Applications CMSMs

CO₂ separation

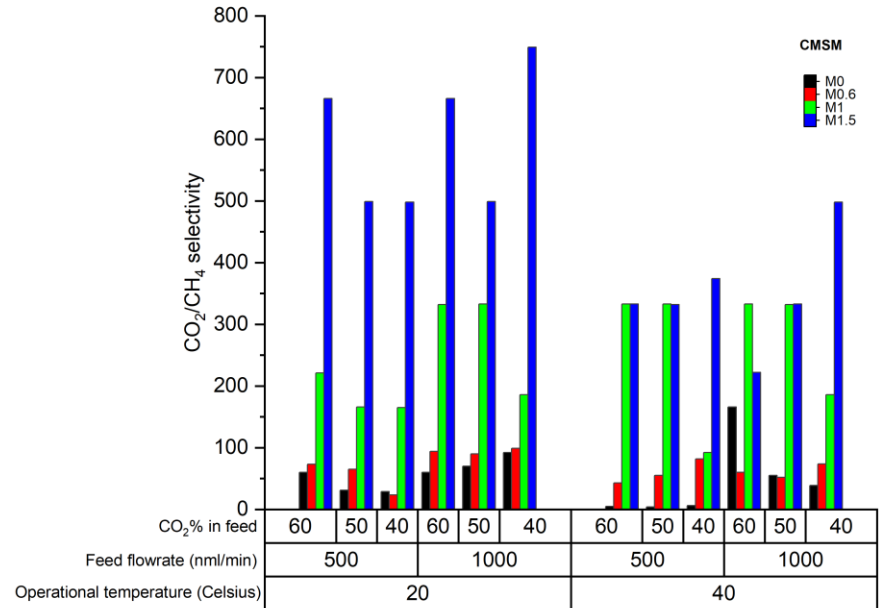
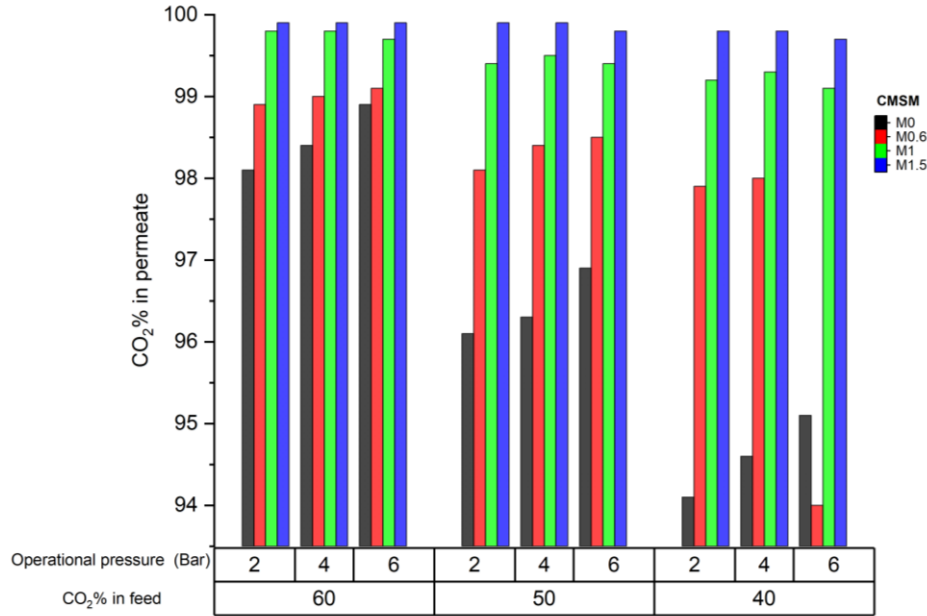
CO₂/CH₄ (Biogas upgrading)



Applications CMSMs

CO₂ separation

CO₂/CH₄ (Biogas upgrading)



Applications CMSMs

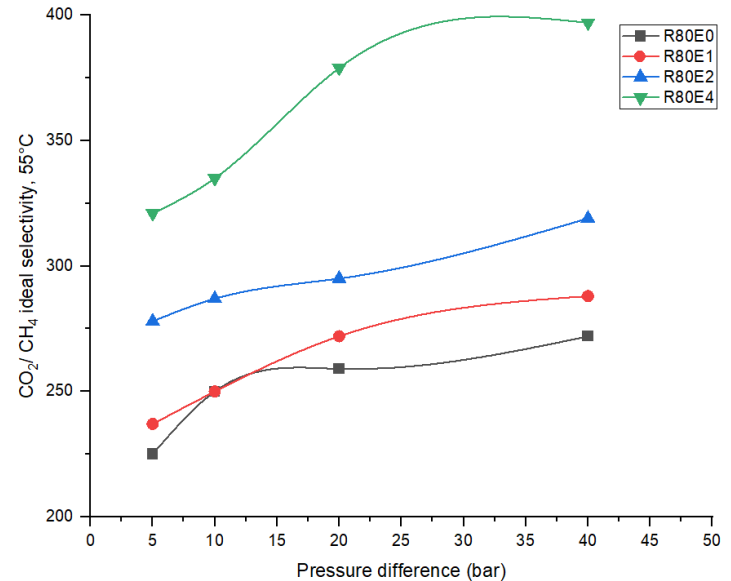
CO₂ separation

CO₂/CH₄ (Natural gas purification)

Increased CO₂/CH₄ ideal selectivity with increasing ethylenediamine content

- Increasing the ideal selectivity dependence on the operational pressure with increasing the ethylenediamine content in the dipping solution because of increased surface diffusion and adsorption sites
- Stable performance at 40 bar operational pressure

Effect of pressure on the CMSMs permselectivity



Applications CMSMs

CO₂ separation

CO₂/CH₄ (Natural gas purification)

Novolac – Ethylenediamine co-polymerization

Ethylenediamine effect:

- Reduction in surface roughness (Ra and Rz)
- Increase of membrane stability at higher temperatures and uniform permeance (less thickness variation)

Membrane	Surface roughness (3D laser microscopy)			
	50x		150 x	
	Ra*	Rz**	Ra	Rz
E 0	286	2021	202	1280
E 0.4	227	1429	69	475
E 1.2	100	734	94	652

* the average surface roughness, **difference between the tallest (peak) and deepest (valley) on the surface.

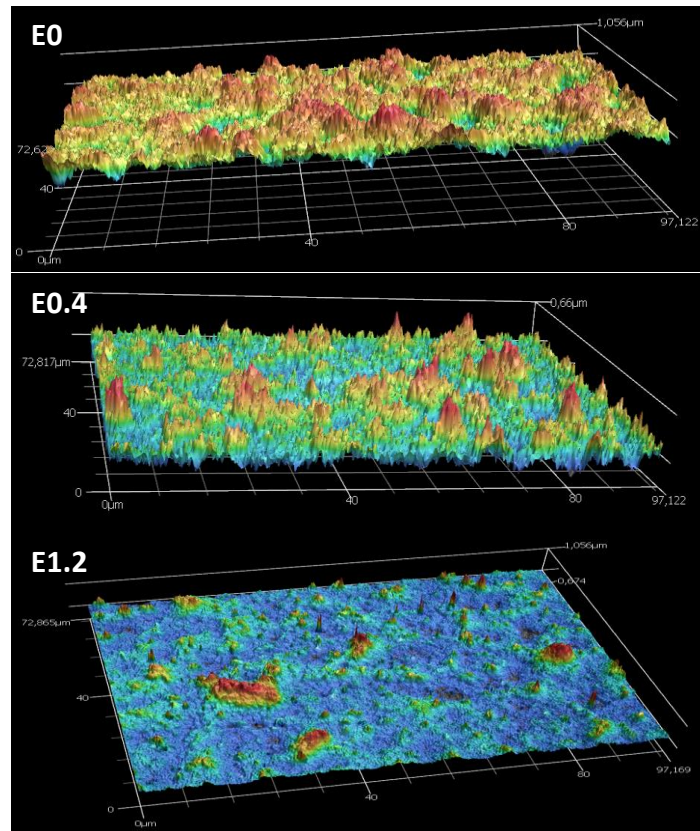
<https://doi.org/10.1016/j.jcou.2022.102378>



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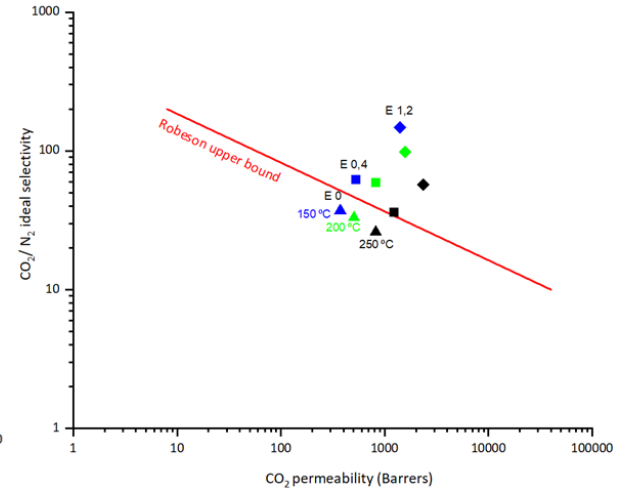
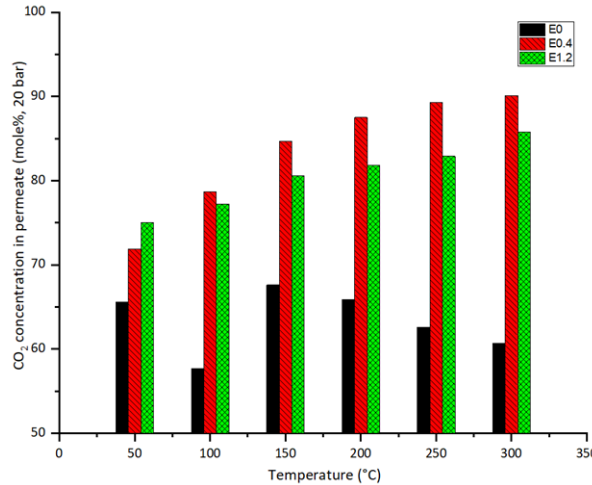
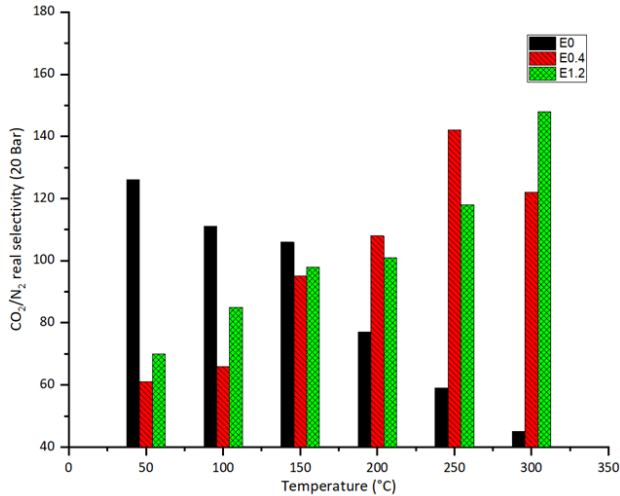
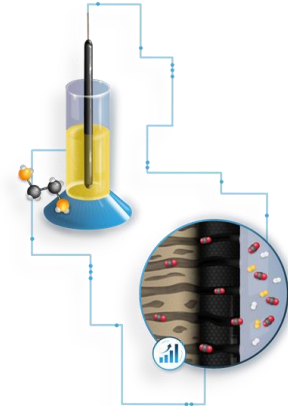
3D laser confocal microscopy



Applications CMSMs

CO₂ separation at elevated pressures and temperatures CO₂/Mix (Steel mill off gas)

- 152 CO₂/N₂ real selectivity at 300 °C
- 92 % CO₂ in the permeate
- Stable performance

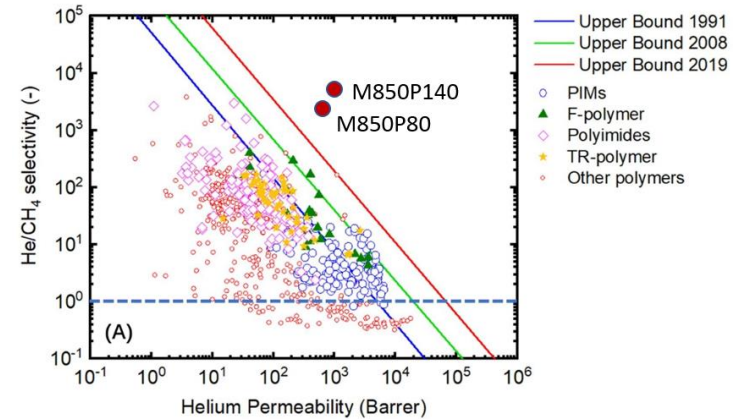
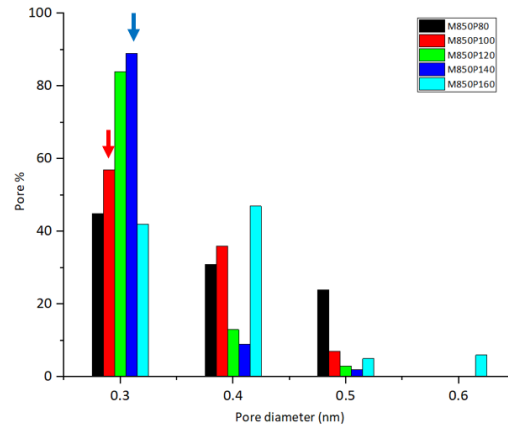
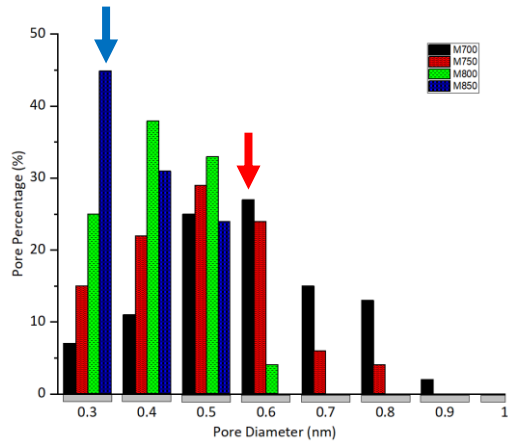
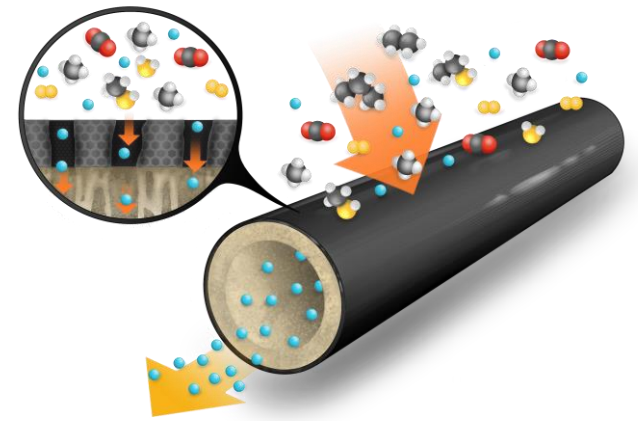


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Applications CMSMs

He separation from natural gas

- Shifting the smaller average pore size with increasing the polymerization temperature but there is an optimum
- CMSM carbonized at 850 °C and polymerized at 140 °C, contains 89% of the pores below 0.3 nm in diameter

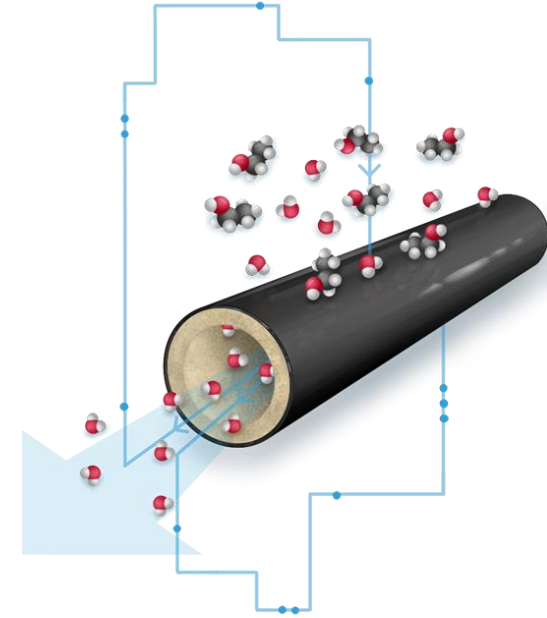
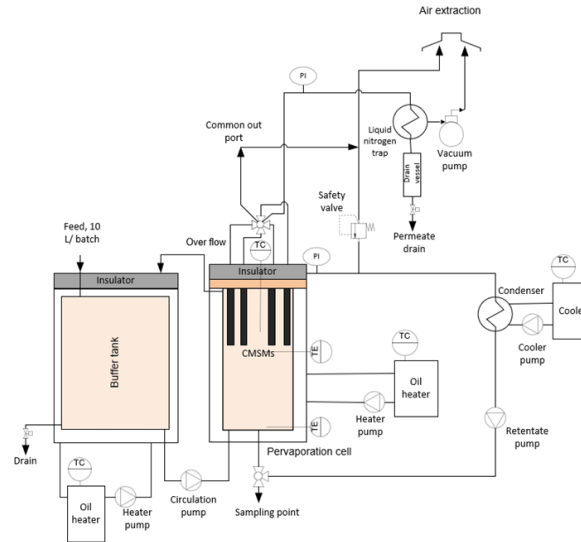
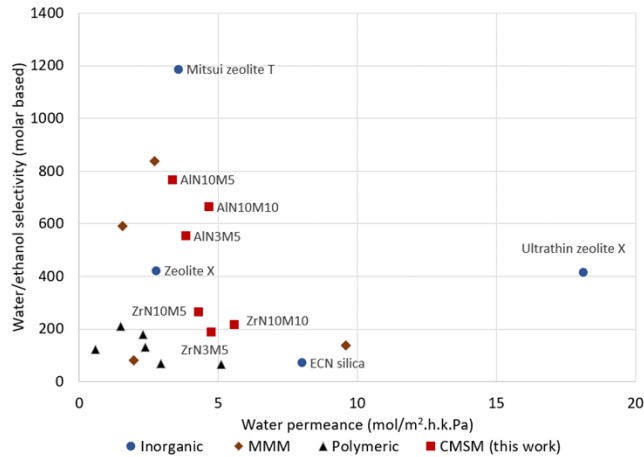


Applications CMSMs

Bioethanol dehydration via pervaporation (2021)

Effect of **support roughness, coating parameters** investigated

The higher **oligomer's molecular weight** resulted in higher water/ethanol selectivities.



<https://doi.org/10.1016/j.cej.2022.134891>

*Thanks for your time and attention
Questions, suggestions are welcome*



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