



Carbon molecular sieves membranes (CMSM) for gas separation and membrane reactors.

- David A. Pacheco Tanaka, Margot A. Llosa Tanco,
- Arash Rahimalimamaghani , Fausto Gallucci.

Solutions to mitigate global warming

- Reduce the emission of greenhouse gases



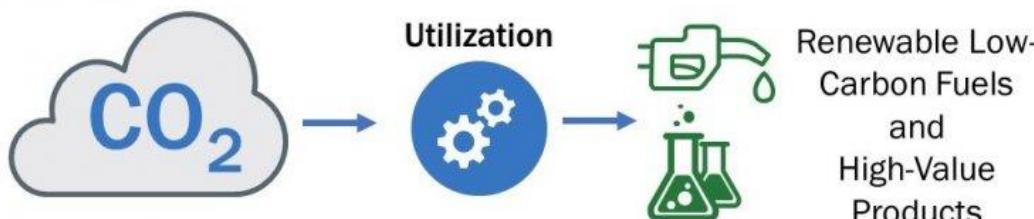
credit : glooly67

- Use of clean fuels

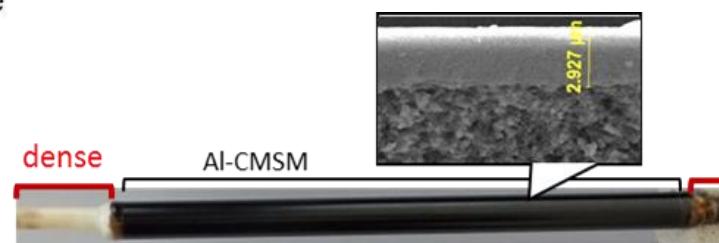


TU/e

- CO₂ capture and utilization

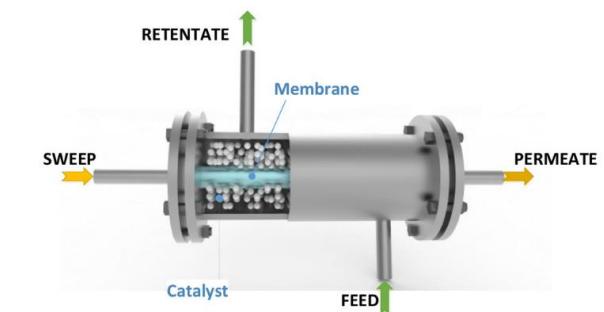


Renewable Low-Carbon Fuels and High-Value Products



tecnal:a

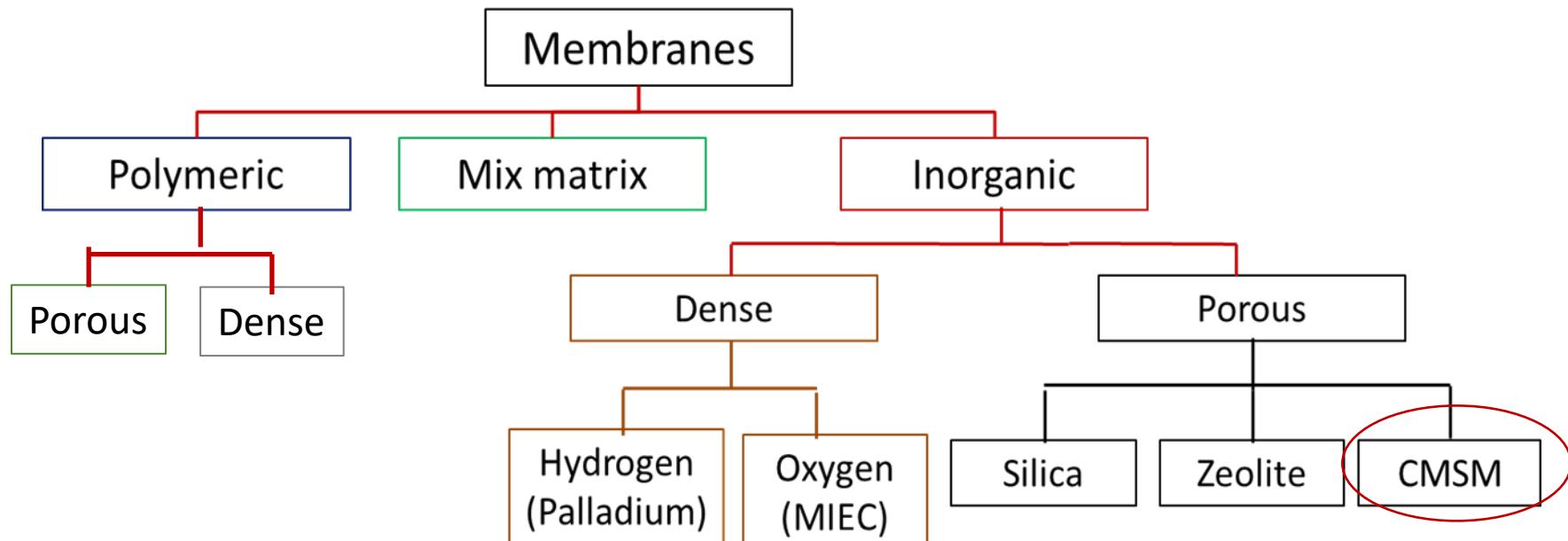
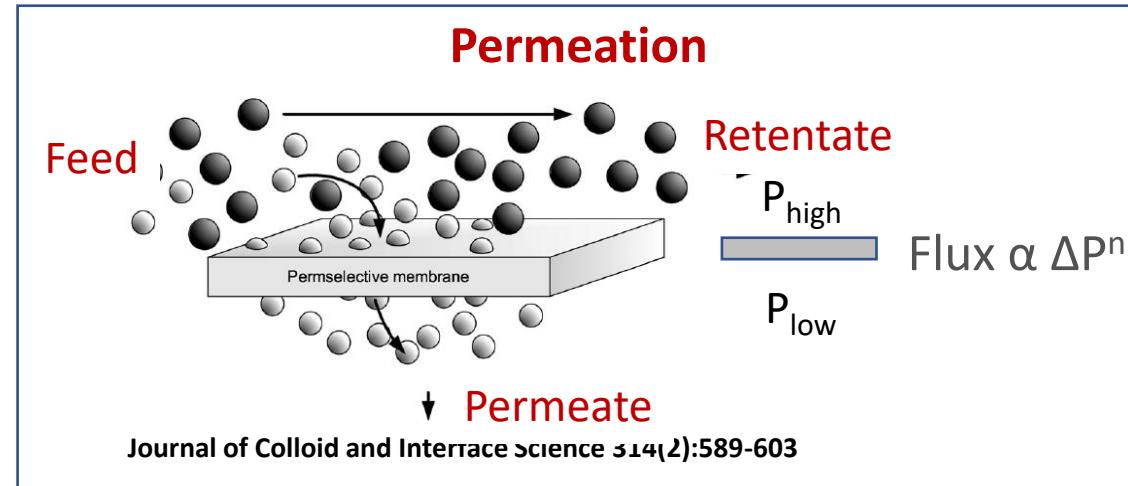
MEMBER OF BASQUE RESEARCH & TECHNOLOGY ALLIANCE



Membranes

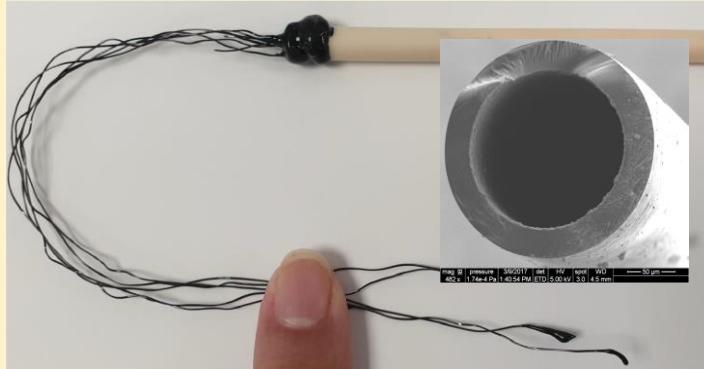
Objective

- ✓ High selectivity
- ✓ High permeation
- ✓ Stable at operation conditions



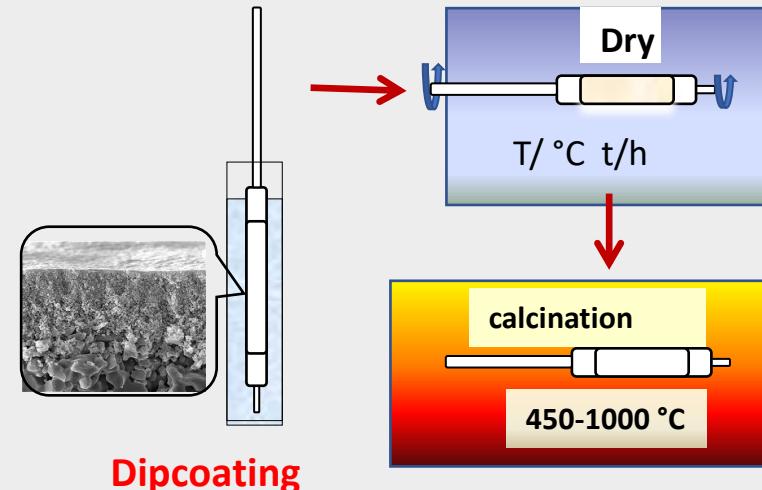
Carbon membranes

-Carbonization thermosetting polymers



Hollow fibre and Self standing are brittle

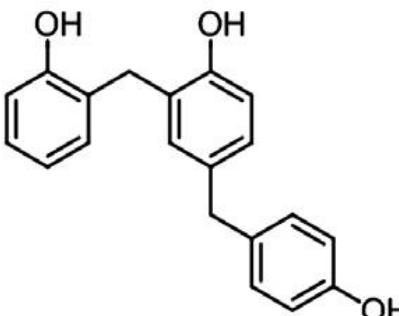
Supported -Alumina-CMSM



Dipping solution

Boehmite nanoparticles	0,8 %
Novolac resin	13,0 %
Formaldehyde	2,0 %
Ethylenediamine	0,6 %
Solvent	NMP

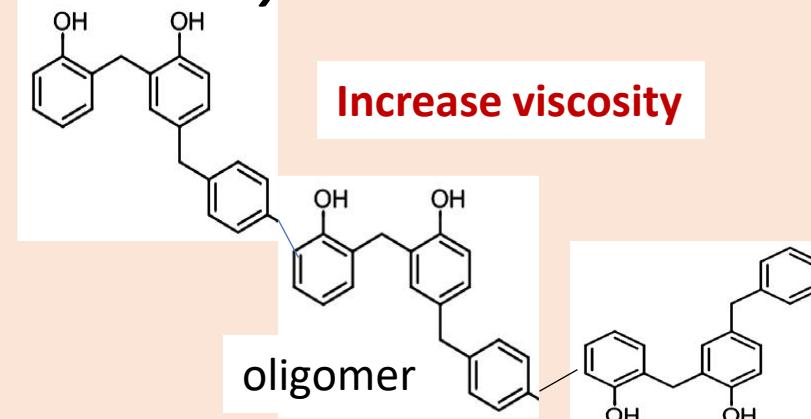
Novolac phenolic



HCHO
(formaldehyde)

Acid: oxalic acid
Basic : KOH
amines

Pre- Polymerization 80 -90 C

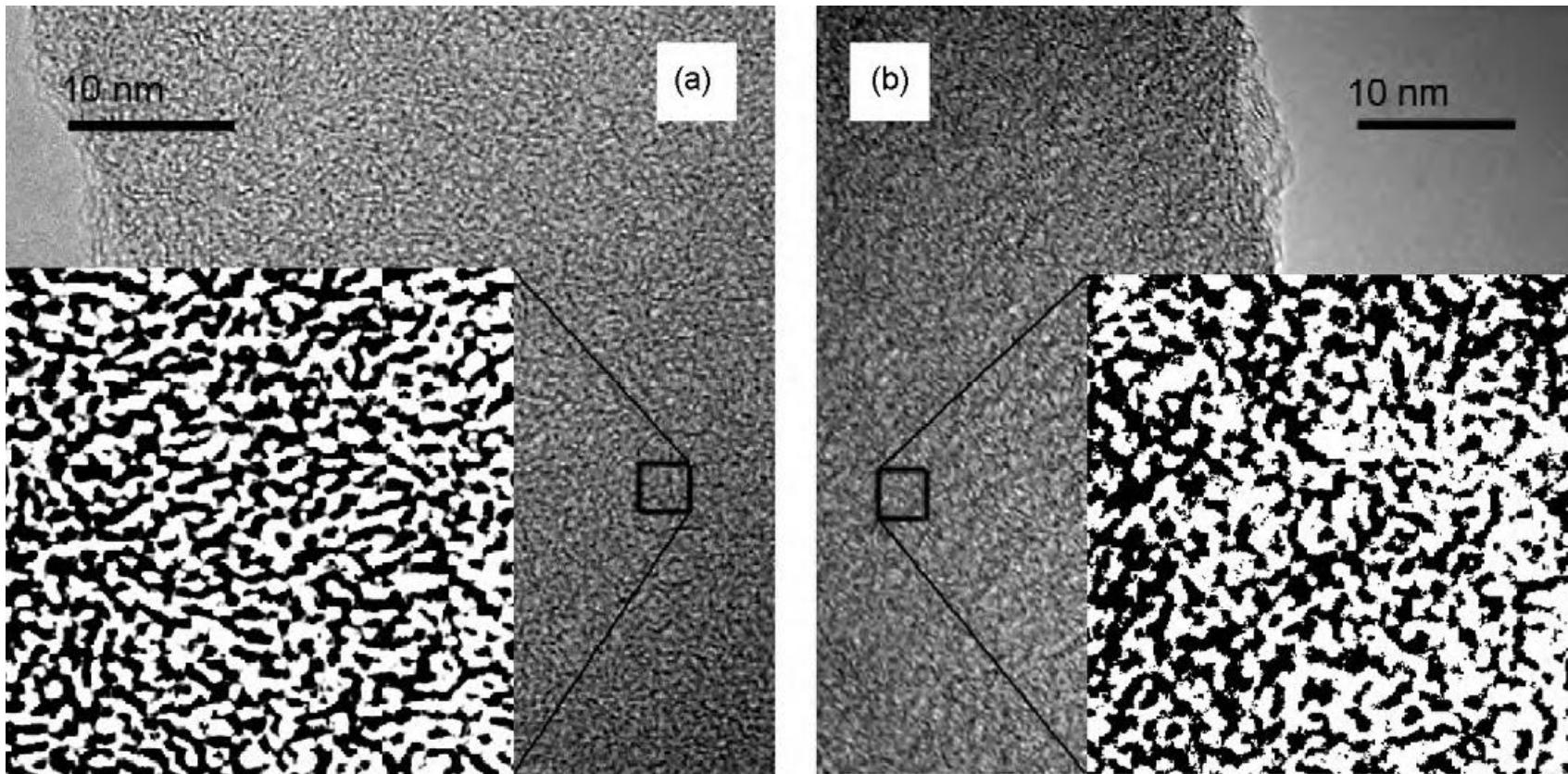


Dip coating

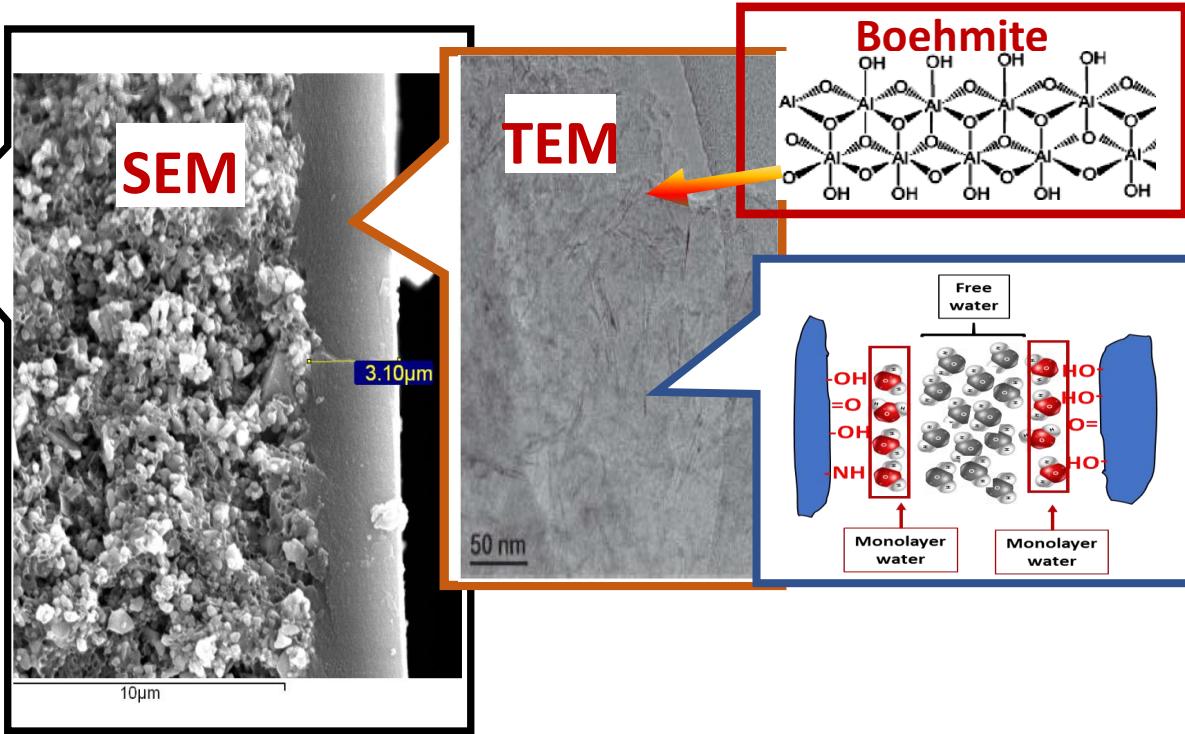
Polymerization on support

Avoid infiltration in support

TEM Carbon Membrane



Chengwen S., Tonghua W., Huawei J., Xiuyue W., Yiming C., Jieshan Q., "Gas separation performance of C/CMS membranes derived from poly(furfuryl alcohol) (PFA) with different chemical structure", *J.Membr. Sci.*, 361, 22-27, **2010**.



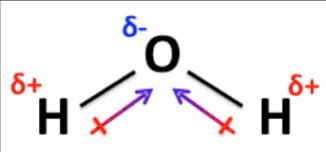
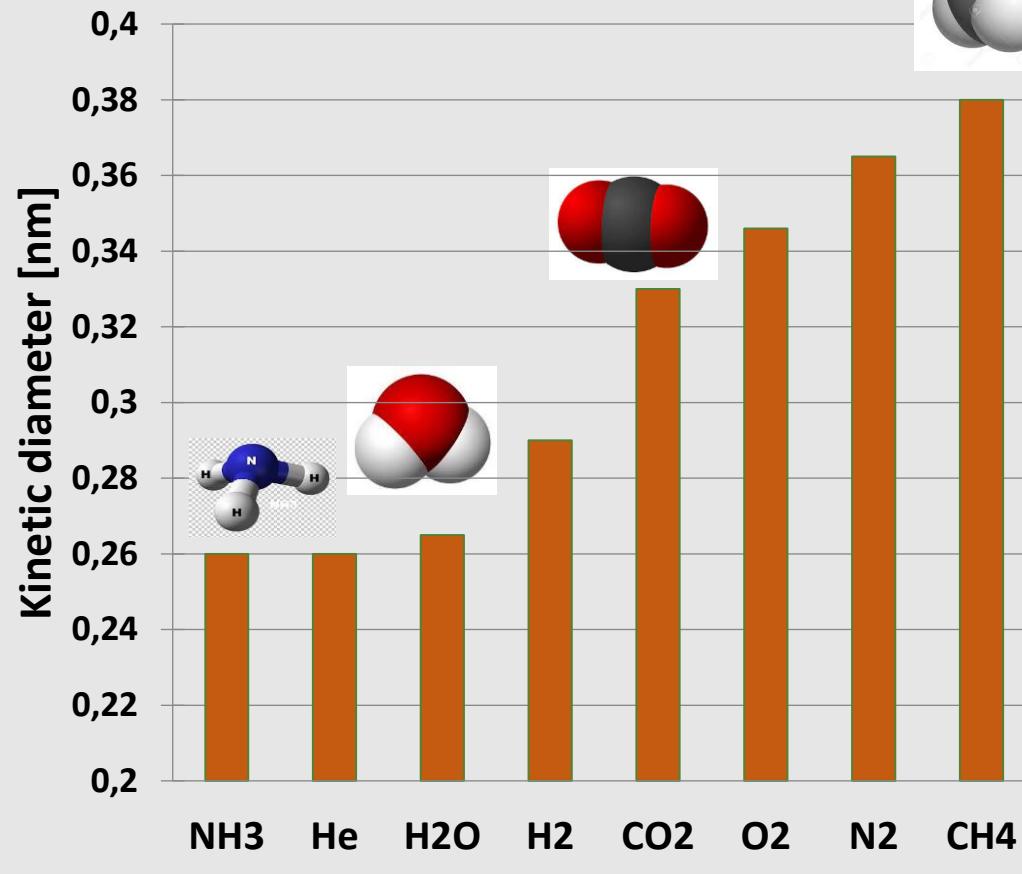
At high temperatures, no water in pores

Knudsen permeation

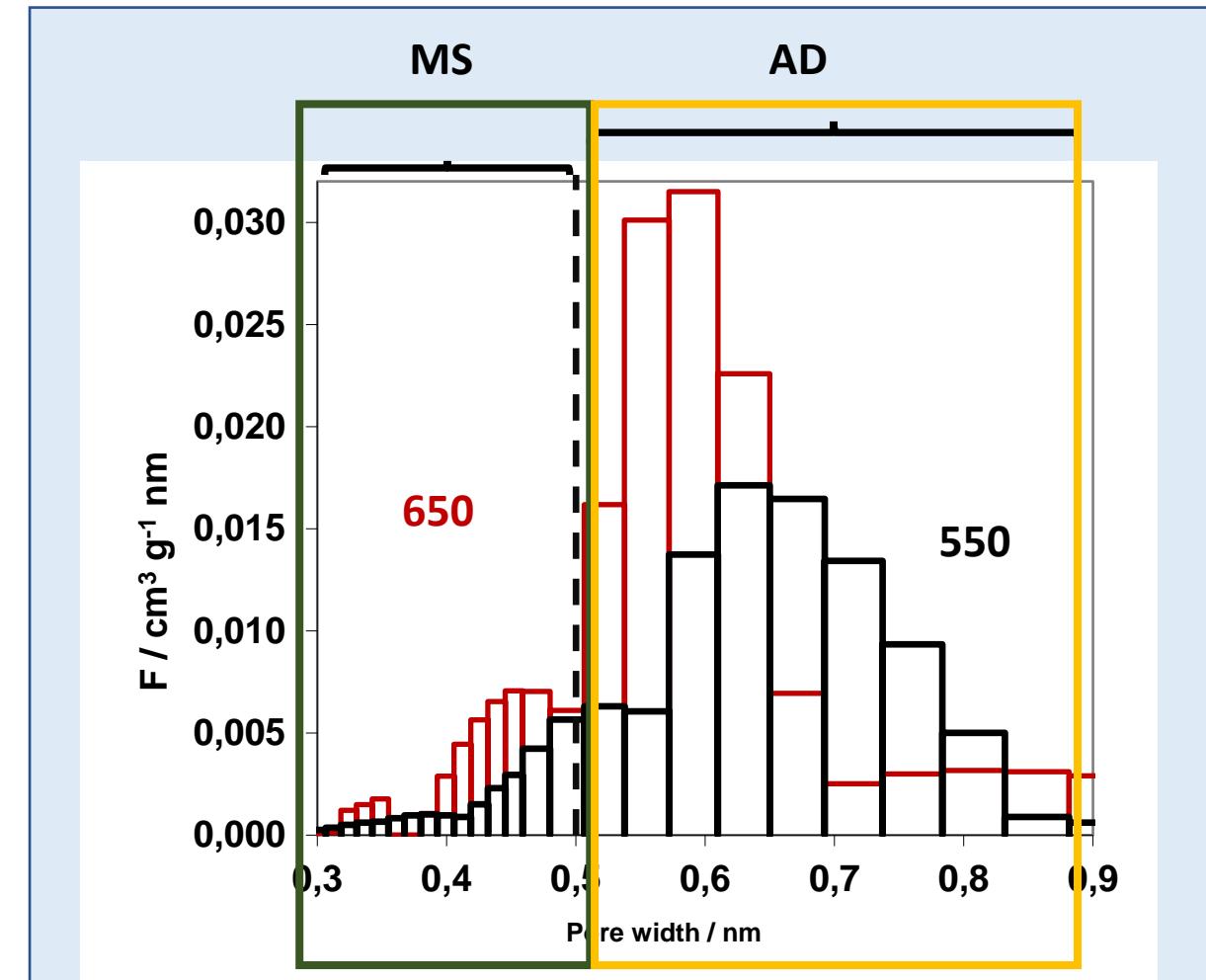
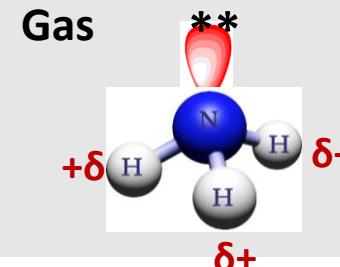
$$\text{flux} = 1/\sqrt{MW}$$

$$\text{flux} \quad H_2 > NH_3$$

$$MW \quad 2 \quad 17$$



NH₃ Dipole, pair electrons (H-bonding)

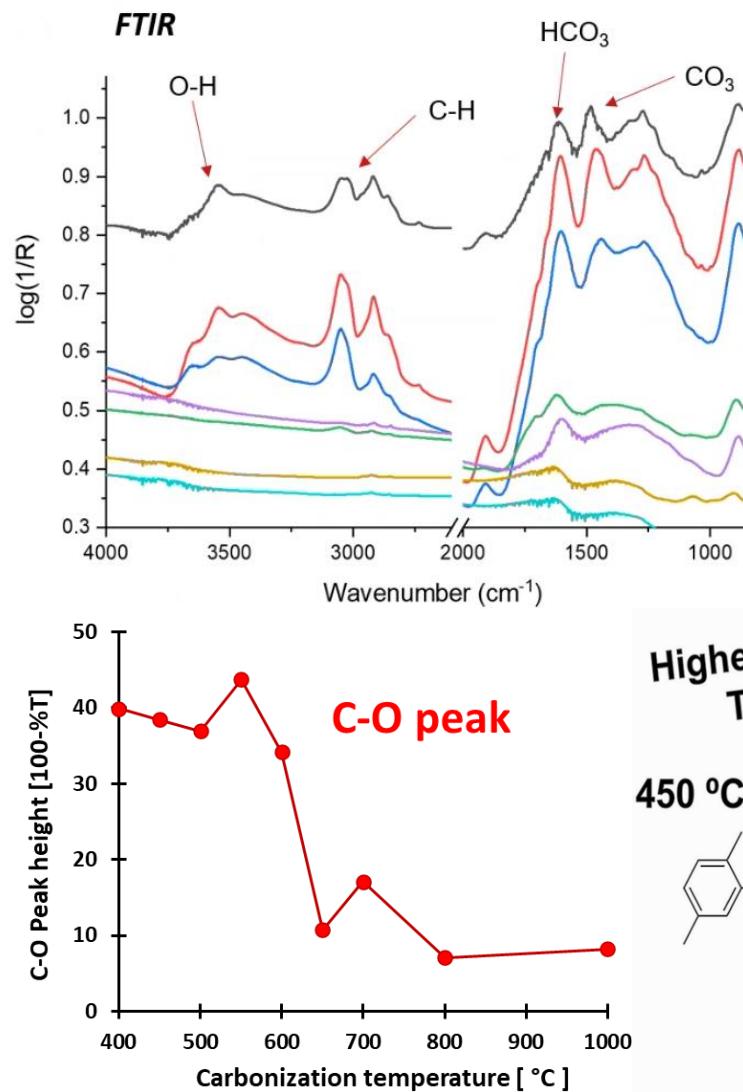


MS
Size separation

Adsorption
Diffusion

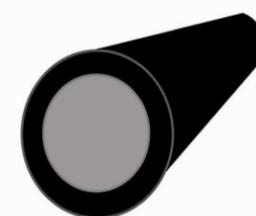
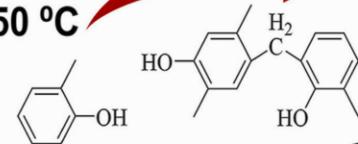
Effect of the temperature of carbonization

FTIR



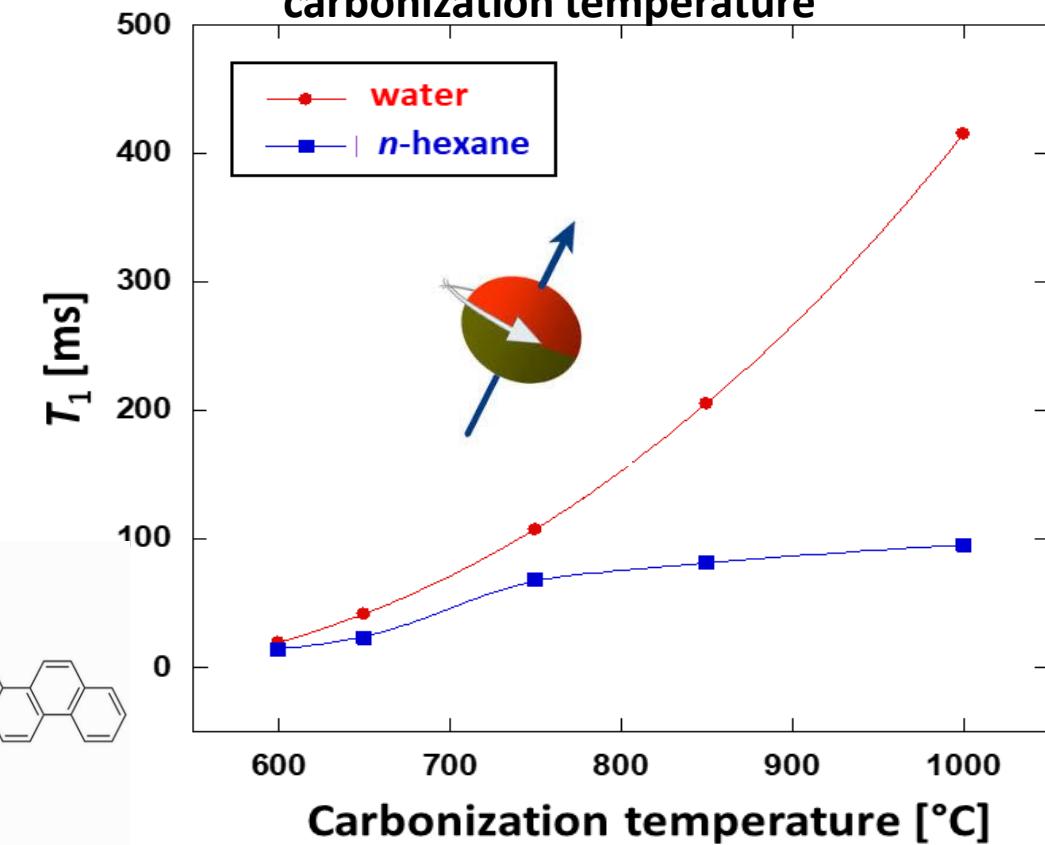
Higher Carbonization
Temperature

450 °C → 750 °C



Proton -NMR

T_1 values of n-hexane and water confined as function of carbonization temperature

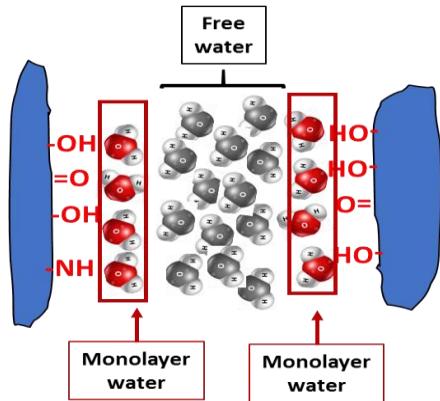


Chemical Engineering Journal 424 (2021) 129313

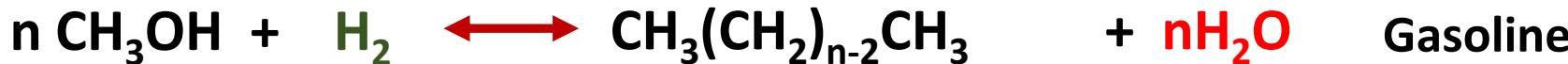
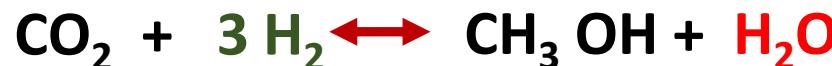
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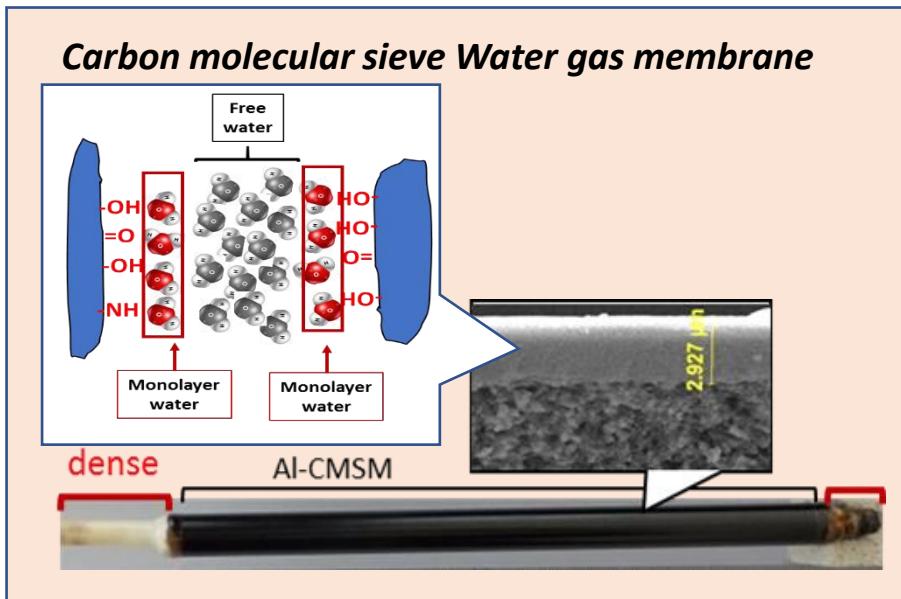
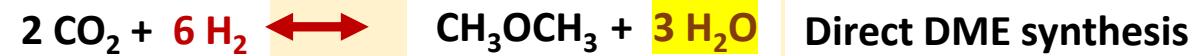
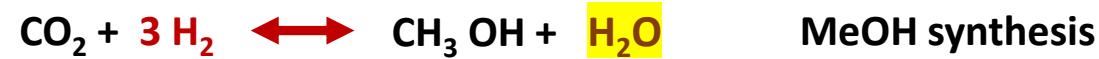
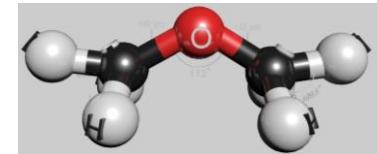


Valorización de CO₂, e-Fuels

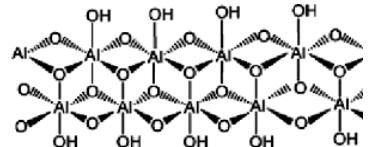




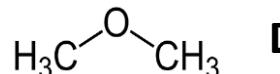
Synthesis of dimethyl ether using membrane reactors



Boehmite



Gas permeation of Al-CMSM containing 0.8 % of boehmite

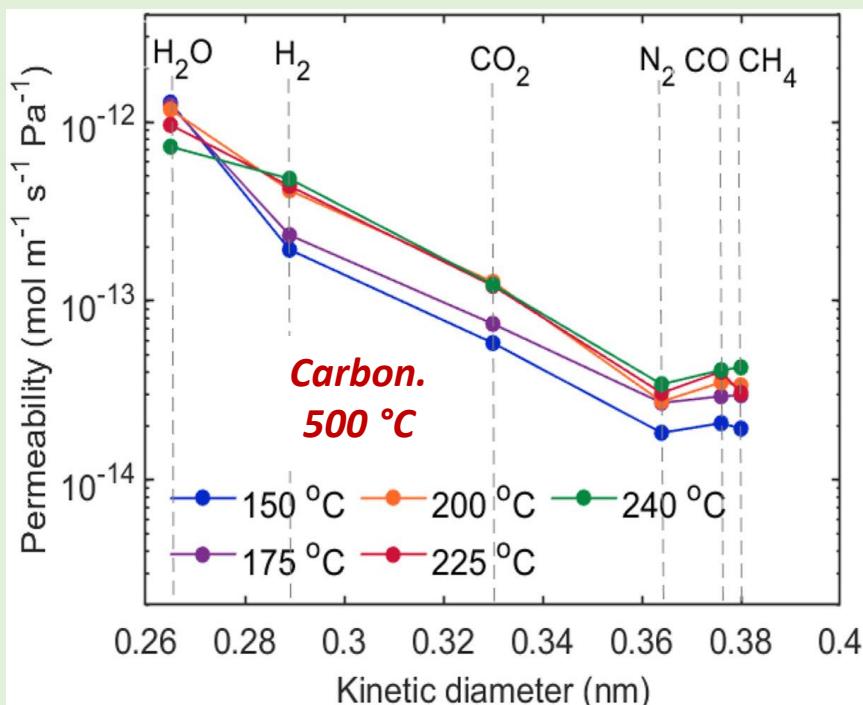


Dimethyl ether



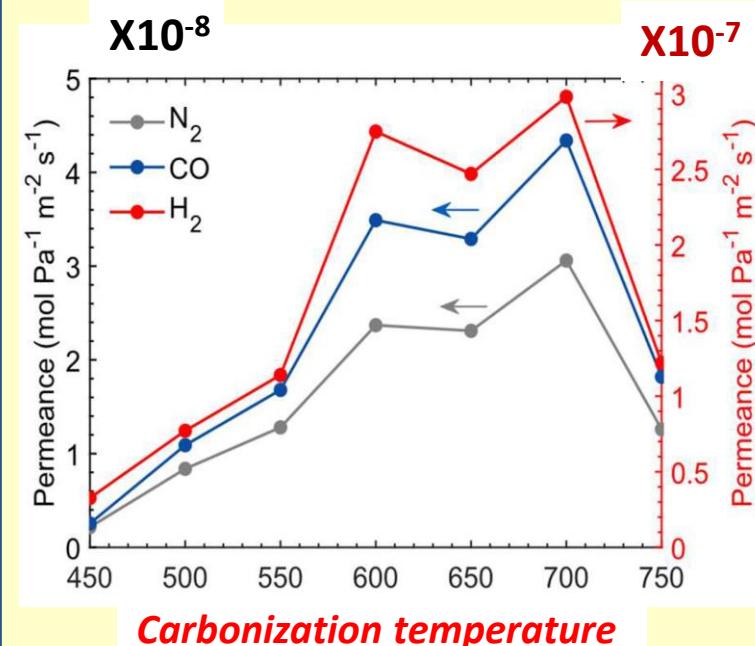
Effect of permeation temperature

Al-CMSM carbonization 500 °C

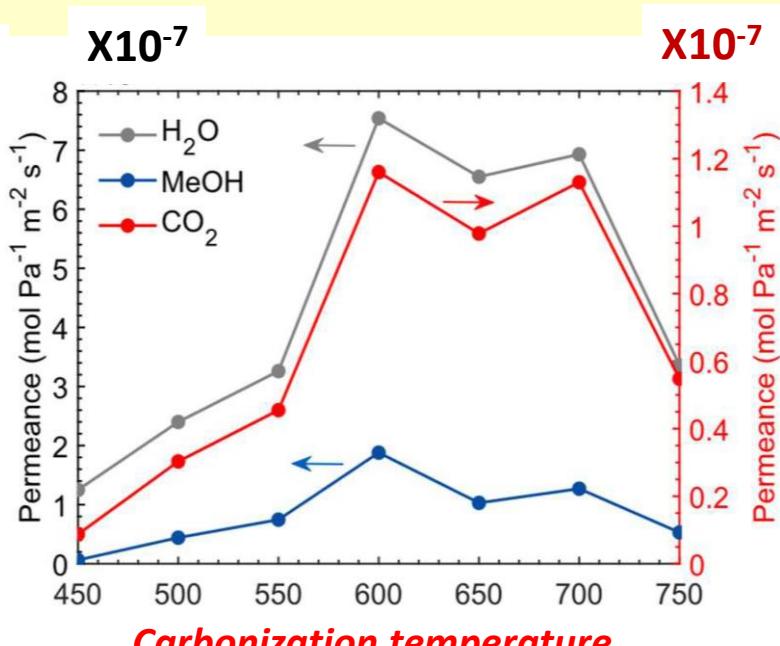


S. Poto et.al. *Int. J. hydrogen energy*
47(2022)11385

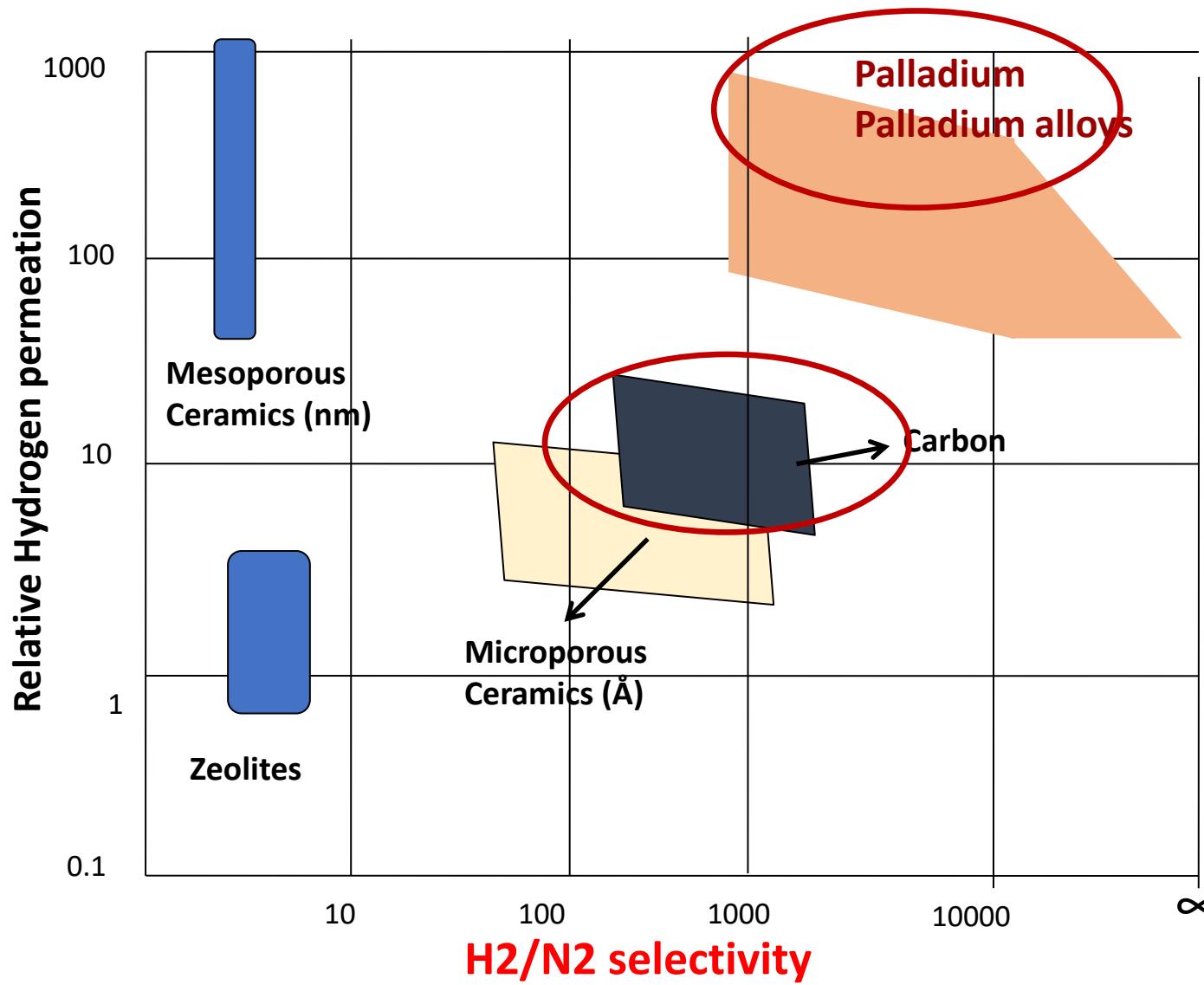
Effect of carbonization temperature (permeation 200 °C)



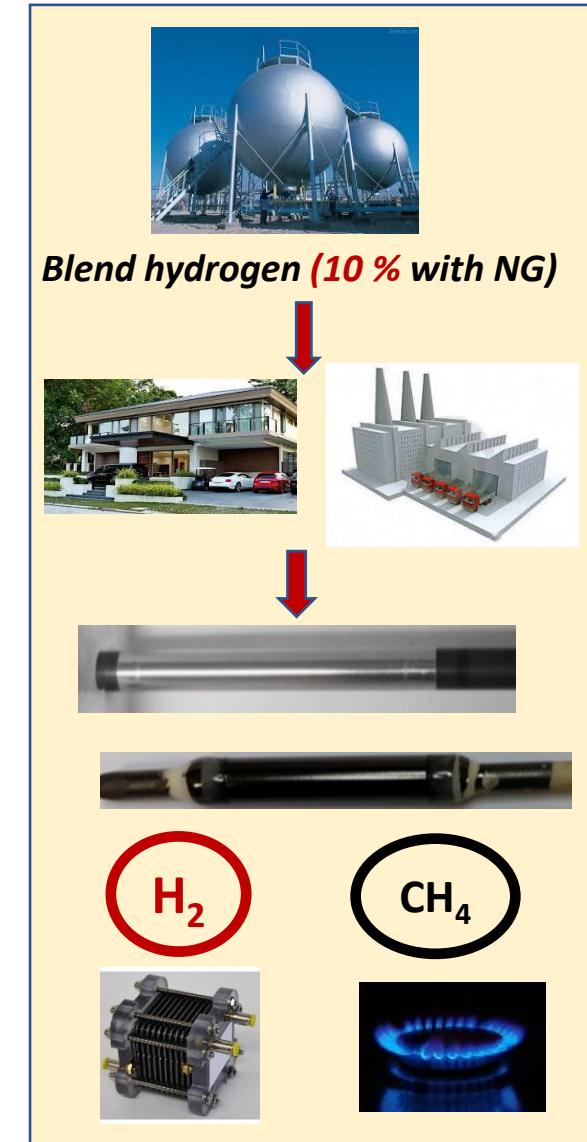
S. Poto et.al. *J. Membr. Sci.* 677(2023)21613



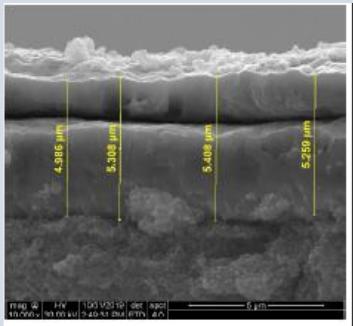
H₂ selective membranes



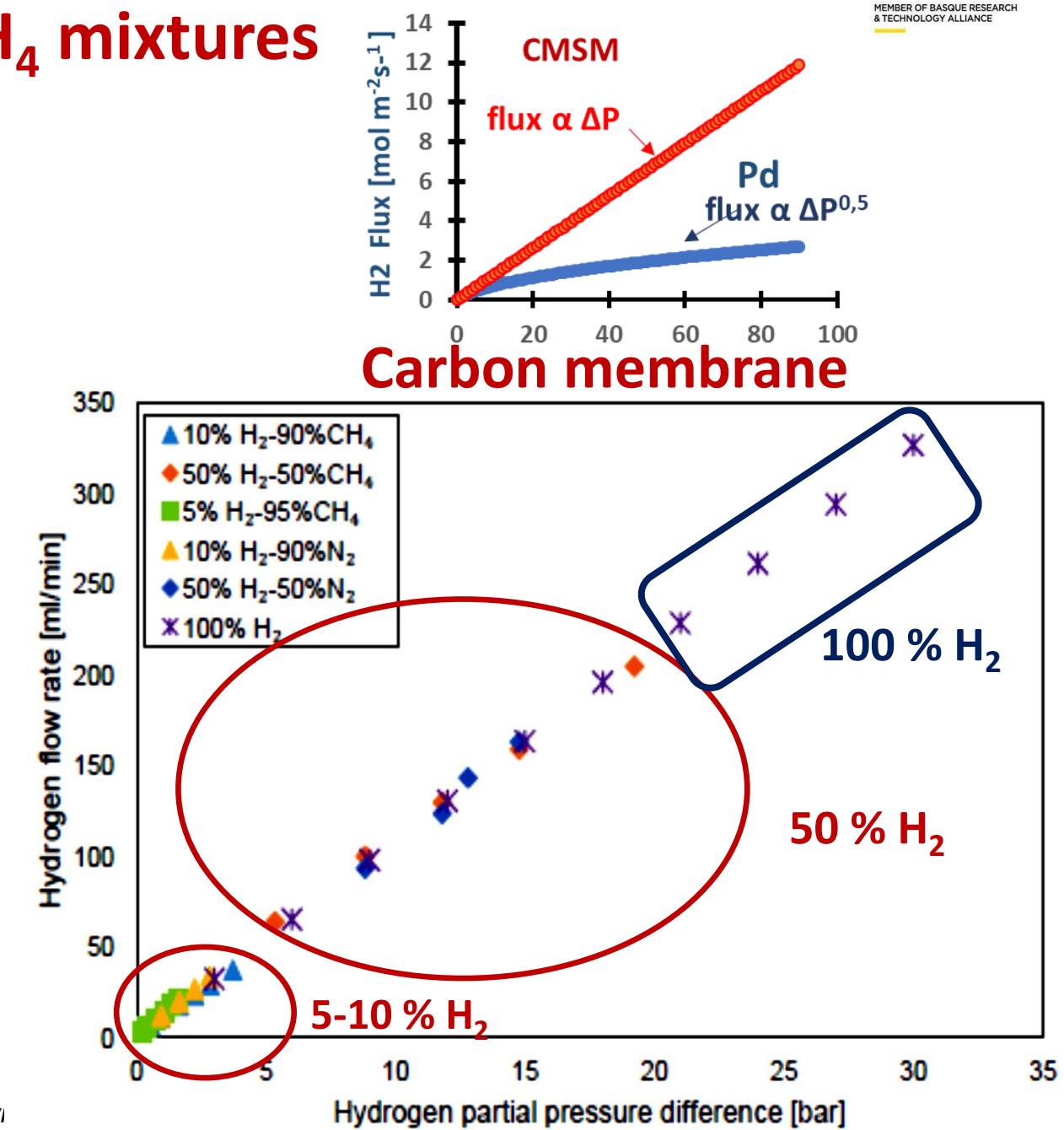
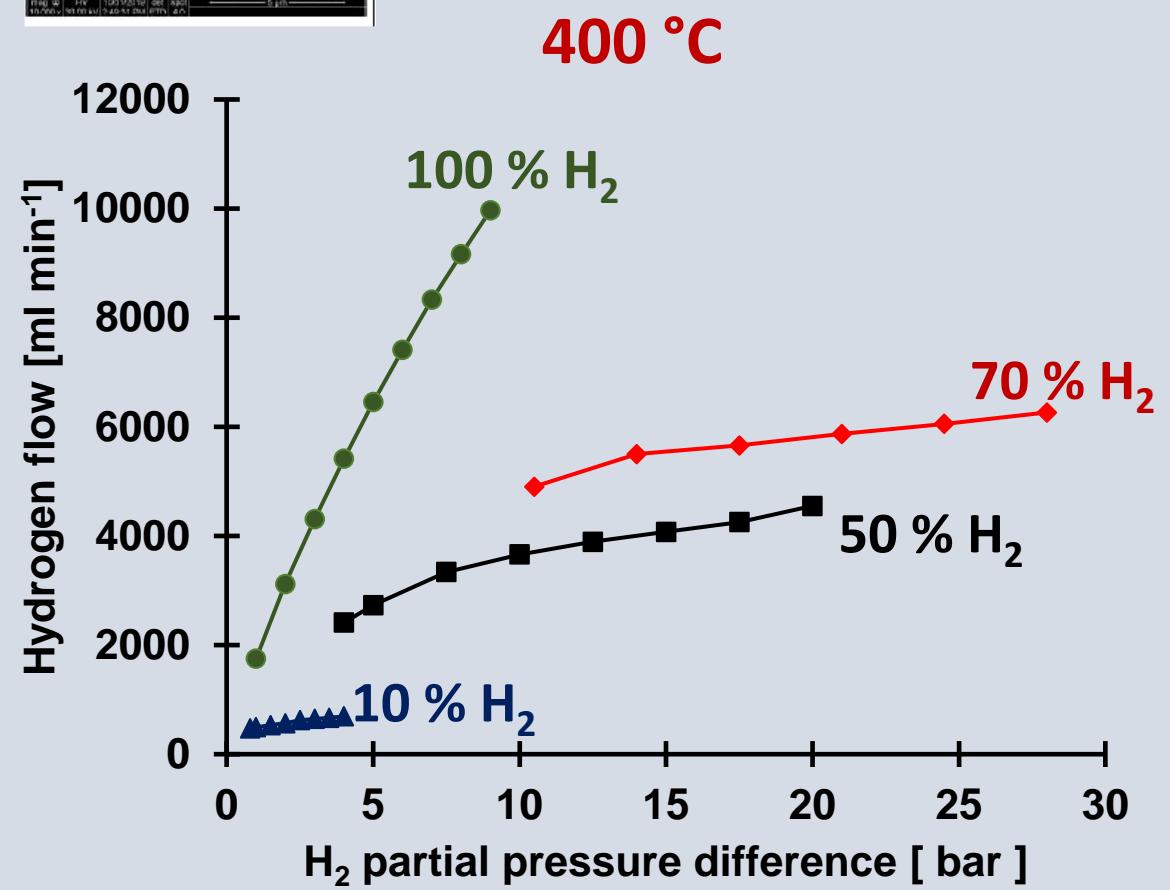
H₂ y Grid



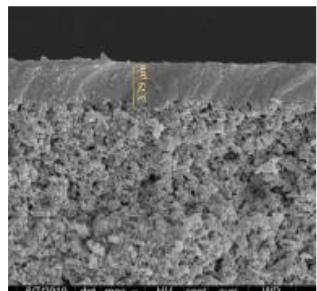
H₂ flow from H₂/CH₄ mixtures



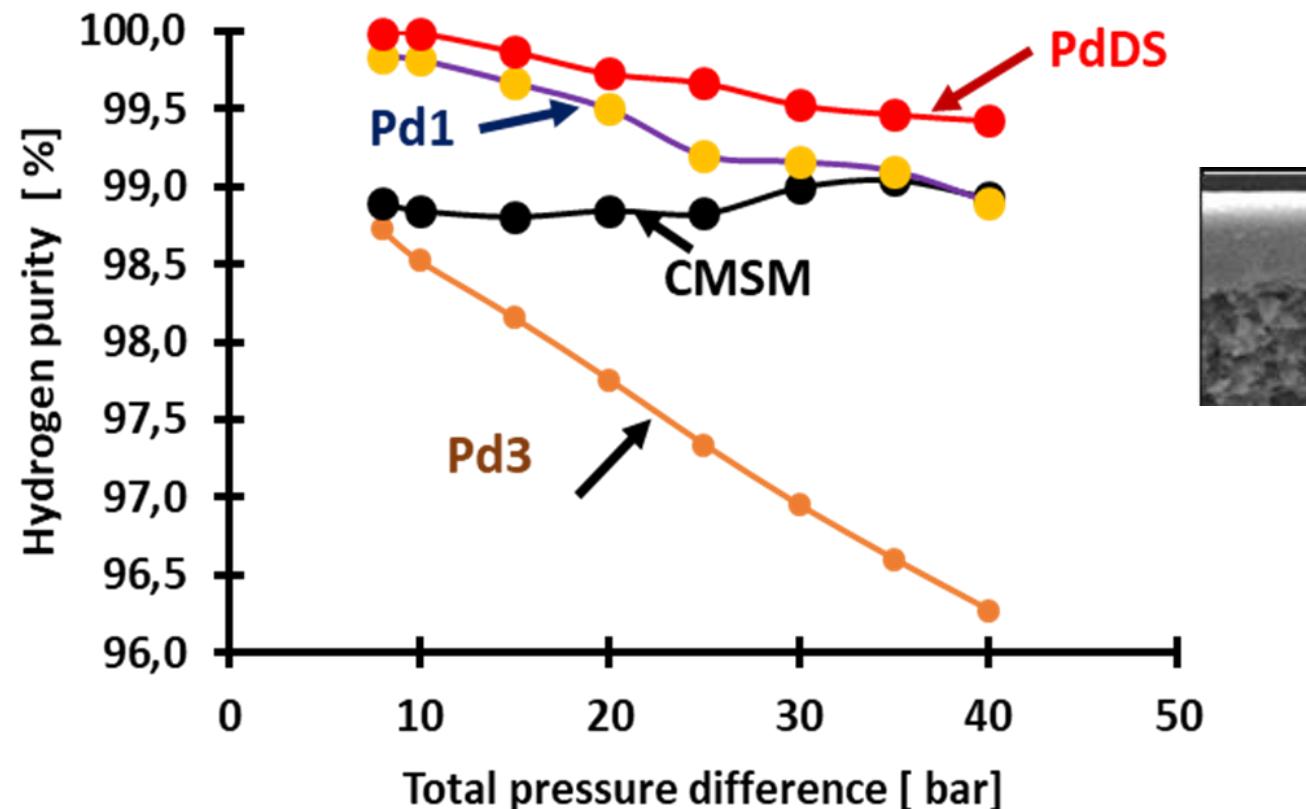
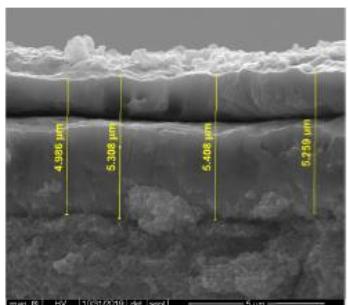
Pd-Ag double skin



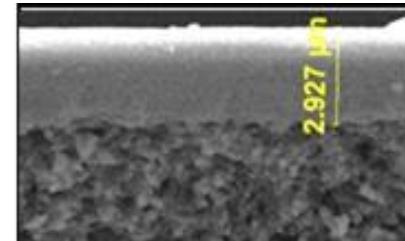
Pd 1 and Pd3



Pd 2 DS



CMSM



Techno-economic evaluation on a hybrid technology for low hydrogen concentration separation and purification from natural gas grid

Maria Nordio ^a, Solomon Assefa Wassie ^a, Martin Van Sint Annaland ^c,
D. Alfredo Pacheco Tanaka ^b, José Luis Viviente Sole ^b, Fausto Gallucci ^{a,*}

Int. J. hydrogen energy 46(2021)23417

Comparison between carbon molecular sieve and Pd-Ag membranes in H₂-CH₄ separation at high pressure

Maria Nordio ^a, Jon Melendez ^b, Martin van Sint Annaland ^c,
D. Alfredo Pacheco Tanaka ^b, Margot Llosa Tanco ^b, Fausto Gallucci ^{a,*}

Int. J. hydrogen energy 45(2020)28876

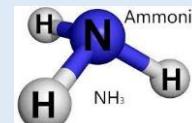
H₂ vs NH₃

H₂



Liquid H₂ is at least 10 times more expensive to produce and store than NH₃ because it requires high pressure and low temperature

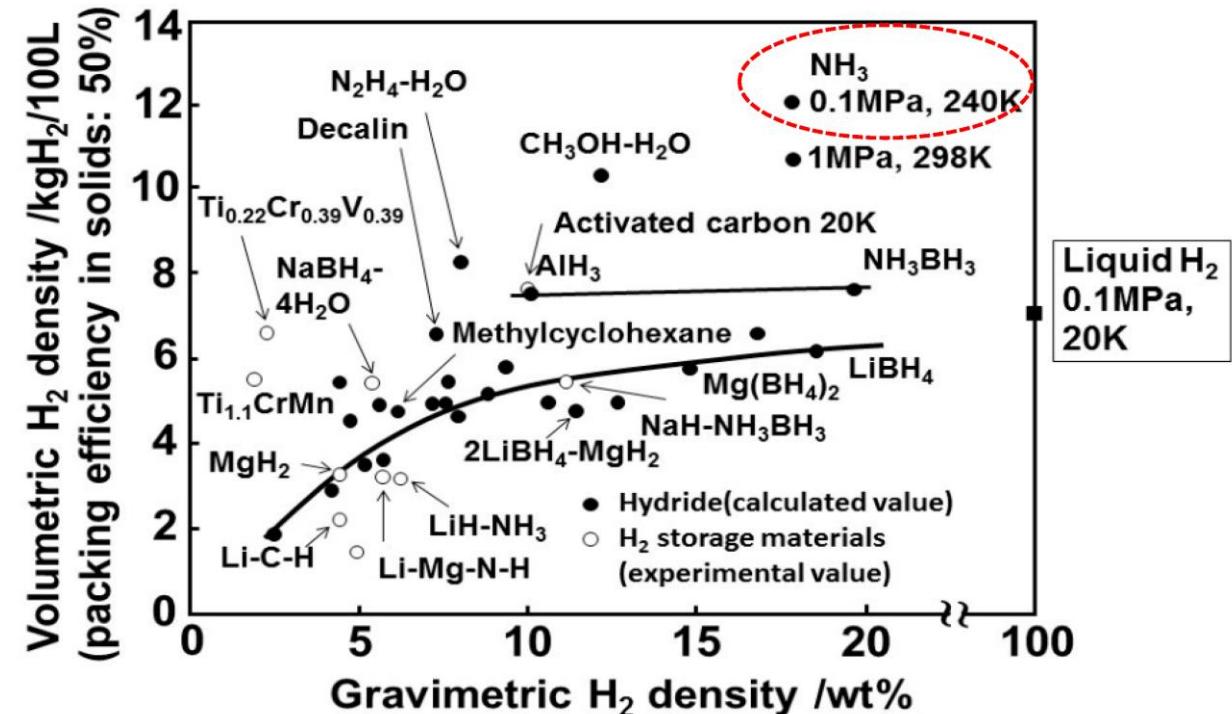
NH₃



Liquid at 10 bar or -33 °C
Ammonia has a supply chain and storage well established

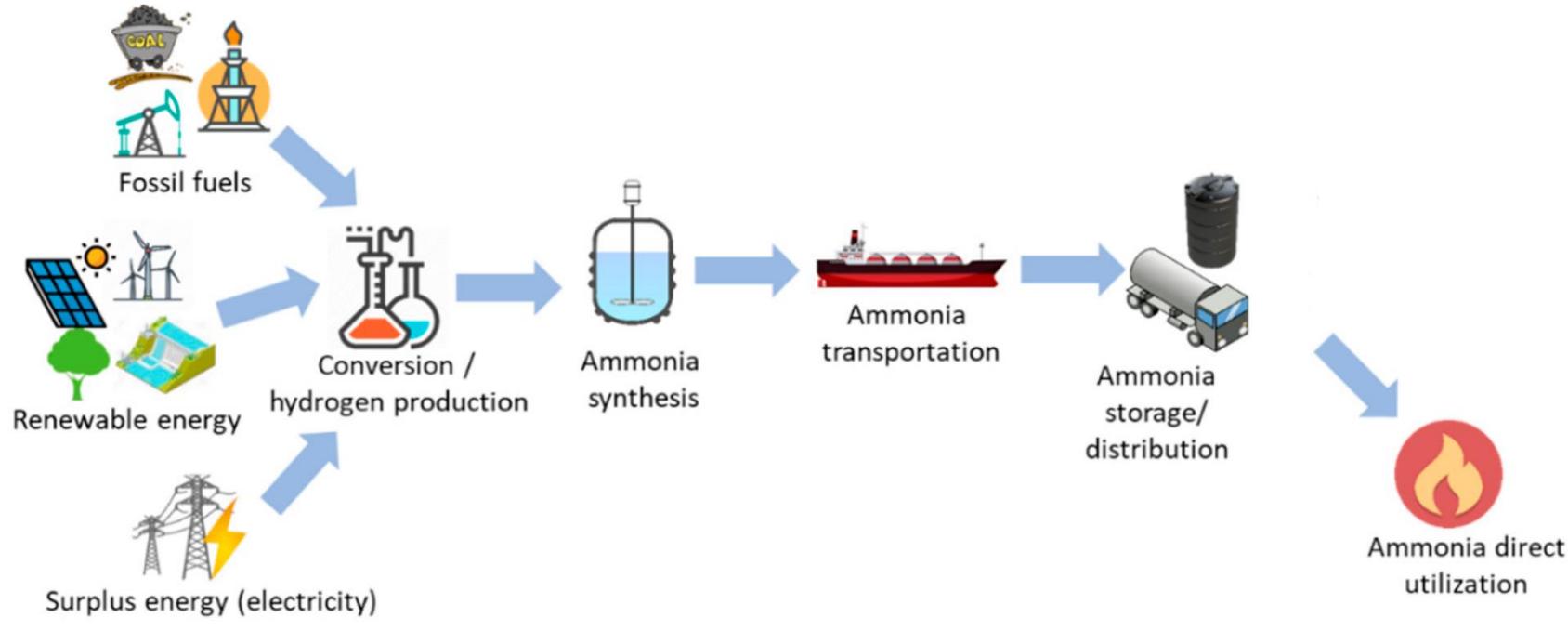
NH₃ was used in internal combustion engines since 1800

The density of hydrogen in hydrogen carriers



Energies 2021, 14(13), 3732

NH₃ as transport and storage of H₂



Descomposition of ammonia

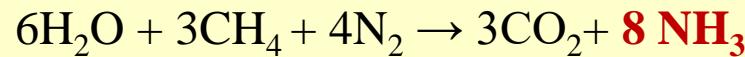


areNH₃a



NH₃ Synthesis

Haber-Bosch (H-B) process (1913)

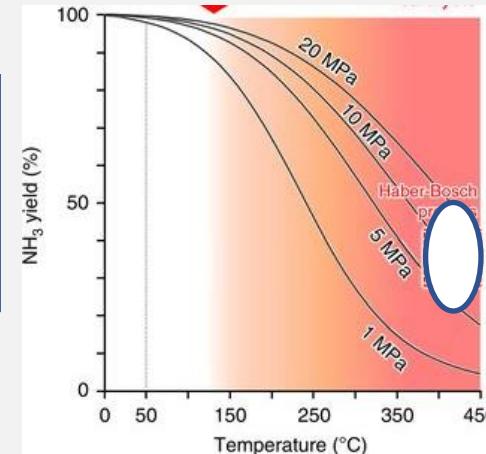
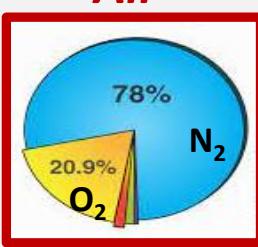


1.8% energy consumed 1.8% CO₂ produced in the world

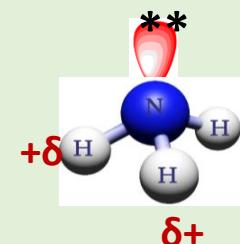
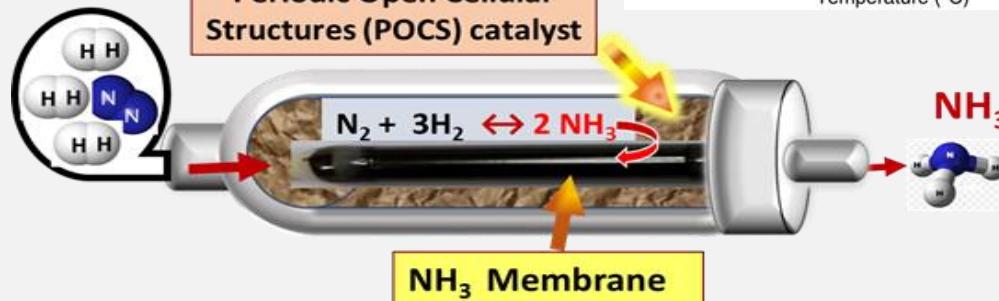
Catalytic Membrane Reactor (CMR)



Air



Periodic Open Cellular Structures (POCS) catalyst



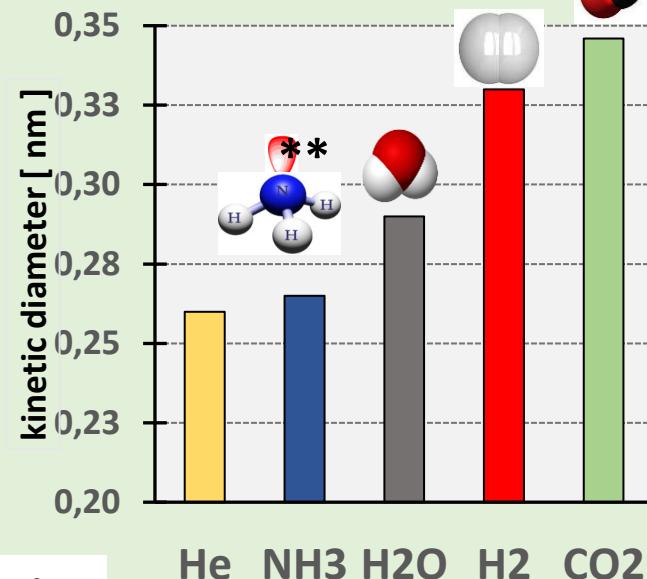
$\delta+$

$\delta-$

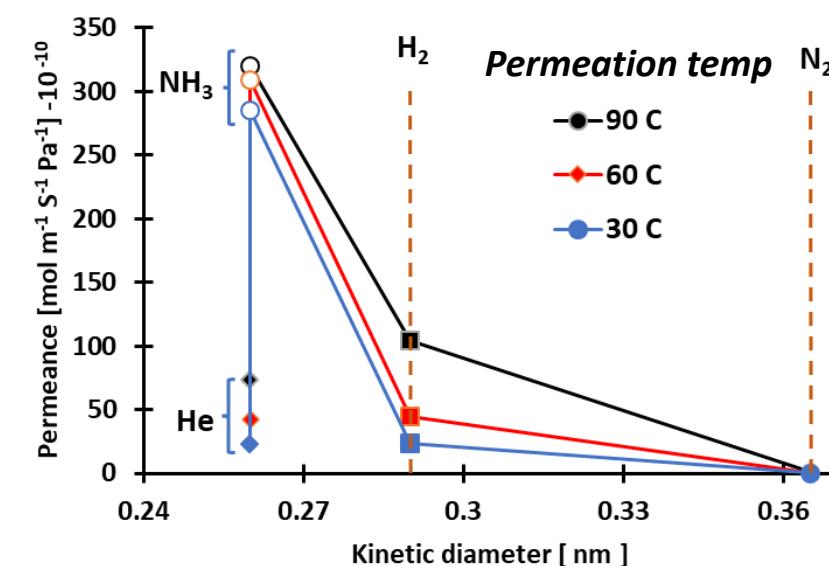
$\delta+$

$\delta+$

NH₃ Dipole, pair electrons
(H-bonding)



Al-CMSM carbonized 500 °C



acknowledgements



Funded by
the European Union



Funded by the European Union under grant agreement No 700355 (Hygrid), 838014 (C2FUEL), 862482 (areNH₃a), 101058565 (Ambher), 101112118 (AndreaH). Views and opinions expressed are however those of the author(s) only and do not necessarily reflect those of the European Union. Neither the European Union nor the granting authority can be held responsible for them



Many thanks